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RECONNAISSANCE OF TELLURIUM RESOURCES IN ARIZONA, COLORADO, NEW MEXICO, AND UTAH

Including Selected Data From Other Western States and Mexico

By F. D. Everett

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RECONNAISSANCE OF TELLURIUM RESOURCES IN ARIZONA, COLORADO, NEW MEXICO, AND UTAH

Including Selected Data From Other Western States and Mexico

by

F. D. Everett 1

ABSTRACT

Thermoelectric and metal alloying applications have changed the status of tellurium from oversupply to undersupply. In a search for new sources of this element in Arizona, Colorado, New Mexico, Utah, and neighboring states, the bureau of Mines analyzed more than 200 samples and obtained the results of tellurium determinations by private companies on more than 500 samples from a wide variety of mineral-bearing deposits.

About 40 minerals containing tellurium have been reported, mainly tellurides of gold, silver, bismuth, mercury, lead, copper, iron, and nickel. The minerals usually occur in small quantities and identification in ores is extremely difficult.

Samples containing more than 0.1 percent tellurium were taken from the Hilltop and Memphis properties in Dona Ana County, N. Mex., and the Silver King property in White Pine County, Nev., where development was chiefly for base metals. The Lone Pine mine in Catron County, N. Mex., and the Bambolla mine near Moctezuma, Sonora, Mexico are potential sources of native tellurium occurring in small but high-grade pods.

Much of the production of tellurium has come as a byproduct of electrolytic refining of copper and lead-zinc ores in Arizona, Colorado, New Mexico, and Utah. Base-metal ore deposits will continue to be the most important primary source of tellurium, although the concentration in ores is nearly always than 10 parts per million. Any significant increase in output of tellurium probably will be related to production and improved efficiency in smelting and refining base-metal ores.

INTRODUCTION

Interest in tellurium has been increasing since World War II, and especially since 1958, because of technical developments in metal alloys and

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Work on manuscript completed July 1962.

thermoelectric devices. The addition of small amounts of tellurium to some metals and metal alloys results in special desirable physical properties. Thermoelectric components using tellurium have favorable characteristics in applications for power generation, heating, and refrigeration. Considerable research is being done in design and application of metal alloys and thermoelectric devices, and industry probably will require much more tellurium than is presently being produced. Future supply will be governed partially by price, which in early 1962 was triple that of 1958. An accelerated demand for tellurium could result in apportioning supply by natural price adjustment according to the value of the uses.

This paper presents the results of a search for possible new sources of tellurium. The investigation was generally confined to the area designated by the Bureau of Mines as Region III, comprising the States of Arizona, Colorado, Nebraska, New Mexico, North Dakota, South Dakota, Utah, and Wyoming, but data available from adjoining areas are included. Mineral deposits investigated for possible new sources of tellurium included copper, lead, zinc, gold, silver, iron ore and pyrite, uranium, sulfur, arsenic, alunite, phosphate rock vanadiferous shale, lignite, and native tellurium. Pulp samples in the Burea of Mines sample files at the Salt Lake City Metallurgy Research Center, together with many samples collected by field investigations, were analyzed for tellurium. Where known, the content of selenium is reported because selenium is frequently associated with tellurium.

ACKNOWLEDGMENTS

Officials of American Smelting and Refining Company, United States Smelting, Refining and Mining Co., International Smelting & Refining Co., Kennecott Copper Corp., The Anaconda Company, American Metal Climax, Inc., Phelps Dodge Corp., and San Manuel Copper Corp. contributed useful information relative to tellurium production and ore deposits.

Acknowledgment is made to S. R. Zimmerly, Director of Research, Kennecott Copper Corp.; J. G. Lecke, General Superintendent of copper refineries, and D. J. Pope, Assistant to the Vice President, mining department, American Smelting and Refining Company; and M. C. Goldsmith, geologist, Minnesota Mining and Manufacturing Co., for special data on tellurium in copper refining and mineral deposits.

HISTORICAL BACKGROUND

Data on historical background, production, uses, and physical characteristics of tellurium have been published. Some of the pertinent data from these publications are repeated here.

Santmyers, R. M. Selenium and Tellurium. BuMines Inf. Circ. 6317, 1930, 23 pp.

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³Fisher, Frank L. Tellurium. Ch. in BuMines Bull. 585, 1960, pp. 851-856.

⁴Hampel, Clifford A. (Ed.), John R. Stone, and Peter E. Caron. Tellurium. Ch. 28 in Rare Metals Handbook. Reinhold Publishing Co., New York, N.Y., 2d ed., 1961, pp. 519-528.

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Tellurium was discovered by Muller von Reichenstein in 1872 while studying a gold-bearing mineral from the Transylvania Mountains in Romania. Klaproth confirmed the presence of the new element and named it tellurium in 1898.

For many years tellurium was a nuisance element in the base-metal smelting and refining industry, but recently the element has acquired new strategic significance because of special thermoelectric characteristics. Tellurium has been produced commercially since 1918 as a byproduct of electrolytic refining of copper and lead. The demand for tellurium was less than 1,000 pounds a year before 1928; annual consumption increased to 22,600 pounds in 1935, to 89,000 pounds in 1940, and since 1950, to more than 100,000 pounds. The price ranged from \$1.50 to \$3.00 a pound between 1918 and 1930, but remained at \$1.75 between 1930 and 1958. The quotation for commercial-grade tellurium in February 1962 was \$6.00 a pound.

Three principal grades of tellurium have been produced--commercial (99.5 plus percent Te), high-purity (99.99 plus percent Te), and semiconductor high-purity (99.999 plus percent Te). Increasing quantities of high-purity tellurium are being used, especially in research on thermoelectricity.

Production

All current tellurium production is derived from the base-metal refining industry. Anode slimes or mud residues contain the gold, silver, platinum, selenium, and tellurium that are carried through concentration, smelting, and electrolytic refining processes. Refineries in the United States recover tellurium--six copper, one lead, and one minor-metal. Copper, nickel-copper, and lead refineries in Canada, and a lead refinery in Peru, recover the major portion of the remaining free-world supply. The U.S.S.R. is known to have a large output of tellurium from the electrolytic refining of metals.

Most base-metal smelters in the United States contribute to the tellurium supply; however, this is accomplished by a system of circulating copper anode slime, copper matte, soda and nitre slags, Cottrell flue dust, baghouse dust, speiss, zinc-leach residue, and lead bullion between smelters and refineries. Base-metal ores, copper concentrates, smelter dusts, and anode slimes from foreign operations, are shipped to plants in the United States for final processing, and these products contribute large but unaccountable quantities of tellurium to the domestic production.

Table 1 lists the production of tellurium in the United States from 1940 through 1961.

Tellurium has been recovered in the past mainly because of its close association with gold, silver, selenium, and platinum in anode muds from electrolytic refining processes. The cost of producing tellurium probably has exceeded its value, especially during the years prior to 1959, when the price for the commercial grade remained virtually stable at \$1.75 a pound.

TABLE 1. - Production statistics of elemental tellurium in the United States, 1940-62, in pounds1

		Primary	Primary producers	Price per
Year	Production	producers	stocks at end	pound
		shipments	of year ²	
1940	85,600	89,000	33,400	\$1.75
1941	224,600	237,700	29,300	1.75
1942	225,000	123,100	134,500	1.75
1943	54,300	48,700	140,900	1.75
1944	61,900	29,600	172,900	1.75
1945	33,500	29,100	176,900	1.75
1946	11,600	48,500	142,100	1.75
1947	60,400	68,300	132,600	1.75
1948	56,900	74,700	114,400	1.75
1949	120,700	68,400	166,600	1.75
1950	107,400	129,900	134,400	1.75
1951	187,100	110,200	147,300	1.75
1952	189,100	94,600	181,100	1.75
1953	70,400	141,200	128,000	1.75
1954	97,100	121,000	103,600	1.75
1955	137,400	166,000	77,000	1.75
1956	223,100	226,100	126,000	1.50-1.75
1957	251,900	210,700	166,000	1.50-1.75
1958	123,200	159,200	131,000	1.65-1.75
1959	176,700	256,900	58,000	³ 1.65-3.00
1960	270,600	227,700	126,000	33.00-4.00
1961	204,500	230,900	64,000	44.00-5.25
1962	264,000	233,000	87,000	55.25-6.00

¹ Bureau of Mines Minerals Yearbooks 1940-62.

Companies producing tellurium in 1962 are listed in table 2.

Most of the electrolytic copper and lead refineries contribute to tellurium production by distributing anode slimes or doré furnace slags to refineries equipped to recover tellurium. Copper and lead smelters also provide flue dusts, baghouse dusts, Cottrell products, or copper matte that are recycled in copper smelting or are distributed to other smelters and refineries where the tellurium and other minor metals are recovered. Leach residue from a zinc refining plant in the midwest contains tellurium which is recovered in a lead refinery.

Some copper and lead smelting and refining companies do not produce tellurium because the recovery costs exceed the value. The recent rise in price should stimulate new interest, and companies may try to improve efficiency in tellurium recovery.

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²Beginning and closing stocks coupled with production and shipments do not balance because imports and exports are not included in table.

³High purity (99.99 plus percent tellurium), \$25.00 per pound.

⁴High-purity tellurium, \$30.00 per pound.

⁵Eng. and Min. J., January-June 1962.

TABLE 2. - Electrolytic copper and lead refineries in the free world that process tellurium from anode slimes

		Location	Refinery
Country	Company	Location	product
United States	American Metal Climax, Inc	Carteret, N.J	Copper.
Do	American Smelting and Refining Co.	Baltimore, Md	Do.
Do	International Smelting & Refining Co.	Perth Amboy, N.J.	Do.
Do	Phelps Dodge Refining Corp.	Laurel Hill, N.Y.	Do.
Do	United States Smelting Refining and Mining Co.	East Chicago, Ind.	Lead.
Do	Penn Rare Metals, Inc	Revere, Pa	Minor metals.
Canada	Canadian Copper Refiners,	Montreal East, Quebec.	Copper.
Do	International Nickel Co. of Canada, Ltd.	Copper Cliff, Ontario.	Copper-nickel.
Peru	Cerro Corp	Oroya	Copper-lead.
West Germany.	Norddeutsche Affinerie	Hamburg	Copper.
Belgium	Society Generale Metalurgie de Hoboken.	Amsterdam	Do.
Japan	Mitsui Mining and Smelting	Takahara	Do.
Do	Mitsubishi Metal Mining Co., Ltd.	Osaka	Do.
Do	Nippon Mining Co	Saganoseki	Do.
Do	do	Hitachi	Do.

Uses

Industry has made little use of tellurium. Before 1930 small quantities were used as crystal detectors in radio sets, as a coloring agent in ceramic and glass products, for photographic tinting, and in chemical products. Presently small quantities are used in alloys of lead, steel, aluminum, tin, copper, and manganese; as a secondary vulcanizing agent in processing rubber; and in insecticides, germicides, and fungicides. Selenium, sulfur, cadmium, and mercury can be substituted for tellurium in some of these applications.

More tellurium will probably be consumed for thermoelectric semiconductors than for all other uses, although in 1960, purchase of tellurium for thermoelectric applications was less than 10 percent of the total consumption. Research has found many uses for thermoelectric devices, based on the Seebeck and Peltier principles. In the Seebeck effect, heat on a junction of special dissimilar materials can be converted into electrical energy, and in the Peltier effect, electric current from an external source passing through such a junction causes heating or cooling, depending on the direction of current through the junction. Additional explanation of these principles is given in appendix A. A good thermoelement must have capacity for high-thermoelectric Power, low thermoconductivity, and low-electrical resistivity. A measure of

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ellue in these factors is called the figure of merit, and some of the compounds that show a high figure of merit are PbTe, Bi₂Te₃, and Bi₂Se₃.⁵

Thermoelectric devices based on these principles have been made to generate small quantities of electricity or to effect refrigeration without moving parts. Efficiency of small electrical generating plants compares favorably to steam or gasoline engines. Efficiency of the electrical-thermo properties is about half that of other popular devices, but the small space requirements and convenience factors are much more attractive. Special applications for thermoelectric devices are numerous, and many are in the research stage.

Possibly nearly all tellurium produced in the future may be consumed in making thermoelectric devices, but because of an apparent limit in output, manufacturers are hesitant to begin making appliances requiring large quantities of tellurium. Several producing companies have formed a research committee to study new applications and promote the consumption of both tellurium and selenium. If tellurium supply remains adequate and the price does not increase greatly, large quantities probably would be used in metal alloys. A small amount of tellurium improves the machinability of some metals and alloys, especially steel.

Special handling procedures must be followed in utilizing tellurium. Workmen exposed to an atmosphere containing as little as 0.001 to 0.01 mg of tellurium per cubic meter of air, have developed obnoxious breaths. This characteristic has deterred several manufactures from using tellurium. Good ventilation, proper use of fume hoods and respirators, and careful handling to avoid contact with dust and materials, are necessary precautions to avoid any disagreeable effects.

SOURCES OF TELLURIUM

The sources of ores contributing to the byproduct tellurium supply cannot always be determined because refineries frequently receive tellurium-bearing products from other refineries and smelters. Ore processors seldom assay for tellurium, and consequently do not know the tellurium content of the crude ores. A general assumption is that most of the tellurium in base-metal ores and in many of the gold-silver ores mined may be recovered. Efforts to improve recoveries have been limited because of the low price. Tellurium has been recovered mainly because its extraction influences the recovery of the higher priced or more-available metals.

Shilliday, Theodore S. Performance of Composite Peltier Junctions of Bi₂Te₃.

J. Appl. Phys., v. 28, 1957, pp. 1035-1042.

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The following companies are members of the Selenium and Tellurium Development Committee (1961): American Metal Climax, Inc.; American Smelting and Refining Company; The Canadian Copper Refiners, Ltd.; International Nickel Co., Inc.; The Anaconda Company; Kennecott Copper Corp.; and United States Smelting, Refining and Mining Co.

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Copper Ores

Sulfide copper ores currently yield an estimated 80 to 85 percent of the total tellurium production. One company official estimated tellurium recovery from copper processing to be about 50 percent. Calculations based on grade and tonnage of anode residues, and assuming 50 percent recovery, indicate copper ores from several deposits in the United States average between 0.5 and 5 ppm of tellurium (0.00005 to 0.0005 percent).

Lead Ores

Base-metal ores containing lead as the principal metal yield an estimated 15 to 20 percent of the recovered tellurium. Ores from lead-zinc deposits contain as much as 1 pound of tellurium per ton, but the average tellurium content of the lead ores, estimated from production and assuming a 50 percent recovery, is between 0.5 and 1.0 ppm (0.00005 to 0.0001 percent).

Gold-Silver Ores

The only tellurium known to be recovered from ores containing gold (and/ or silver) is a product of base-metal smelting and refining. Gold ores processed by amalgamation, cyanidation, or gravity concentration are not known to have yielded commercial production of tellurium except where tailings may have been shipped to base-metal smelters. Tellurium in gold ores must be volatilized by roasting before the gold fraction can be recovered by amalgamation or cyanidation. Frequently the dusts and Cottrell precipitates from the roasting operations are returned to the concentrating circuit to prevent pollution of the atmosphere.

Silver tellurides are found alone with gold, and associated with basemetal deposits; only silver tellurides processed with base metals have contributed to the commercial supply.

SOME ECONOMIC ASPECTS OF TELLURIUM PRODUCTION

In the past, production of tellurium has exceeded demand. With the prospect of new applications, demand should exceed production. According to production specialists of American Smelting and Refining Co., the maximum production that could be achieved by improving the efficiency in the copper- and lead-refining industries in the Western Hemisphere is between 500,000 and 750,000 pounds of tellurium a year. Any additional needs must be acquired from new sources. The most likely source for new supply will be base-metal processing plants not contributing, at least to full potential, to the present production.

Two native tellurium prospects were being explored in 1961--one in Sonora, Mexico, and the other in New Mexico. It appears probable that the two Properties could contribute 10,000 to 20,000 pounds of tellurium a year, with exploration and mining comprising the major costs. However, it is unlikely

Page 520 of work cited in footnote 4.

that either property will become an important contributor to the tellurium supply unless the price for tellurium rises above \$20 per pound for commercial grade. At this price a small operation producing 10 to 20 tons of ore a day probably could make a profit on ore containing 2 to 3 pounds of native tellurium per ton (0.1 to 0.15 percent Te).

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Ores containing gold and silver tellurides were mined in Colorado during 1961 in the Cripple Creek and Gold Hill districts in Teller and Boulder Counties, respectively. The Golden Cycle Corp. processed the Cripple Creek ores in the Carlton mill by flotation and cyanidation, until the mill was closed in January 1962 because of inadequate supply of ores from the mines. The mill was designed for a capacity of about 1,000 tons of ore a day, but the rate of milling in 1961 was below 300 tons a day. About 90 percent of the gold was recovered in a flotation concentrate representing 5 percent of the mill feed. A sample of the flotation concentrate assayed 0.07 percent; a sample of the flotation tailing, 0.003 percent Te. The flotation concentrate was passed through a fluosolids reactor, where the ore was roasted and most of the tellurium was volatilized. The tellurium-bearing gases and dust were scrubbed, and the resulting solids returned to the processing circuit. The tellurium ultimately was discarded in the tailing. A company official estimated that at 300 tons a day (12 to 15 tons of flotation concentrates), the operation could recover about 8 pounds of tellurium. As the rate of gold mining and the price of tellurium has been low in the past years, the company was unable to justify tellurium-recovery facilities. Custom processing charges for gold ores scaled upward from \$3.00 a ton. Apparently profitable recovery of tellurium at the Carlton mill would require a much higher price for tellurium and an assured ore supply much greater than 300 tons a day.

TELLURIUM MINERALS

Tellurium is the sixth rarest of the 92 elements found in nature, and the average content in surface rocks is about 0.002 ppm. The physical appearance of tellurium is more metallic than other members of the oxygen family, to which it belongs, but chemically, tellurium resembles selenium and sulfur.

Tellurium most commonly occurs in combination with gold, silver, bismuth, lead, mercury, or copper. Microscopic and chemical determinations are required to identify many of the tellurium minerals. Most of the minerals containing tellurium are listed in table 3.

Telluride minerals range in appearance from silvery white with columnar prismatic cleavage to dark, massive grains. Hardness ranges between 2 and 3.5 Mohs, and specific gravity is 6.2.

Tellurium is a poor conductor of electricity. It has a metallic appearance and melts at 450° C to a dark liquid resembling lead, and boils at 990° C. Chemically, tellurium is similar to sulfur and selenium. Bivalent tellurium ion forms compounds with active metals, and covalent tellurium forms compounds with other elements. Tellurium minerals are mainly stable. Some compounds of tellurium are toxic but generally are harmless to the human body, being absorbed through the skin and by ingestion and inhalation of dust, fumes, and

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vapors. Some forms of tellurium are characterized by an offensive odor similar to garlic. No toxic effects resulting from mining tellurium-bearing ores have been reported.

TABLE 3. - Tellurium-bearing minerals1

Mineral	Approximate	Te, percent
	composition	
Native tellurium	Te	92-100
Tellurite	TeO	78-80
Paratellurite ²	TeO2	78-80
Calaverite	AuTeg	56-58
Petzite	AuAg, Te,	32-35
Sylvanite	(Au, Ag)Te	60-63
Krennerite	(Au, Ag)Te	56-59
Muthmannite	(Ag,Au)Te	39-48
Hessite	Ag ₂ Te	35-38
Stuetzite	Ag ₄ Te(?)	22-23
Empressite	AgTe	53-55
Rickardite	Cu ₄ Te ₃	59-60
Weissite	Cu ₅ Te ₃	53-55
Vulcanite	CuTe	65-67
Nagyagite	Pb ₅ Au(Te,Sb) ₄ S ₅₋₈	17-19
Altaite	PbTe	37-39
Tapalpite	Ag ₃ Bi(S,Te) ₃	20-24
Tetradymite	Bi ₂ (Te ₂ ,S) ₃	33-37
Tellurobismuthite	Bi ₂ Te ₃	45-49
Gruenlingite (also Grunlingite)	Bi ₄ TeS ₃	12-13
Joseite	Bi ₃ TeS(?)	14-16
Wehrlite	Bi ₃ Te ₂ (?)	29-35
Montanite Oruetite	Bi ₂ O ₃ TeO ₃ ·2H ₂ O Bi ₈ TeS ₄	17-20
Coloradoite	HgTe	38-40
Melonite	NiTe ₂	80-82
Emmonsite	Iron hydrated telluride	59-60
Durdenite	Fe ₂ (TeO ₃) ₄ H ₂ O	42-57
Mackayite	Fe(TeO ₃) · xH ₂ O	-
Blakeite	Ferric telluride	-

See footnotes at end of table.

TABLE 3. - Tellurium-bearing minerals1 (Con.)

Mineral	Approximate composition	Te, percent
Colusite Goldfieldite Teineite Arsenotellurite ³ Deinkamite ³ Hedleyite ³ Chicolovaite ³ Rubisite ³ Frobergite ³ Goldschmidtite ³ Tellurium selenide ³ Tellurous hydrogen ³	PtTe ₃ (?) Cu ₃ (Sn,Te,V,As)S ₄ 5Cu ₂ S.(Sb,As,Bi) ₂ (S,Te) ₃ Cu(Te,SO ₄) ₂ H ₂ O Te ₂ As ₂ S ₇ PbTeO ₃ Bi ₇ Te ₃ Bi ₂ TeS ₂ (Te,Se) ₃ FeTe ₂ (Au,Ag) ₂ Te ₅ SeTe	65(?) 0.4-1.3 - 29.6-40. 17.6-20. - 65-97

Dana's System of Mineralogy, John Wiley & Sons, Inc., New York, 7th ed., v. 1, 1944, and v. 2, 1951.

Switzer, George. Paratellurite, a New Mineral From Mexico. Am. Mineralogist, v. 45, November-December 1960, p. 1272.

³Sindeyeva, N. D. Mineralogy, Types of Deposits and Basic Features of Selenium and Tellurium Geochemistry. Acad. of Sci. of U.S.S.R., Inst. of Miner., Geochem. and Crystallochem. of Rare Elements, Acad. of Sci., U.S.S.R. Press, Moscow, 1959, 444 pp.

INVESTIGATION OF TELLURIUM PROPERTIES

A search for potential new sources of tellurium was undertaken in Arizona, Colorado, New Mexico, North Dakota, South Dakota, Wyoming, Utah, and Nevada. Field samples were collected from mineral prospects, mines, mills, smelters, and refineries, and were analyzed for tellurium (appendix B). Data provided by an international mining, smelting, and refining company, involving analysis of several hundred samples from properties in the United States, also were studied (appendix C).

Varieties of mineral occurrences were sampled, with the objective of finding deposits of tellurium. Tellurium minerals mentioned in publications were reviewed for information on sources (appendix D). Samples from mineral deposits containing copper, lead, zinc, gold, silver, iron, uranium, sulfur, arsenic, alunite, phosphate rock, vanadiferous shales, fluorite, and other elements and mineral commodities were analyzed for tellurium. Samples representative of a large mineral production were taken where feasible. High-grade ores or highly mineralized rocks were selected from some deposits with the view of finding indications of tellurium. Samples of concentrates and smelter products also were collected. Smelting companies disclosed the origin of shipments known to contain tellurium-bearing ores processed at their plants.

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Some base-metal shipments from mines in northern Mexico to smelters in Arizona and Texas contained relatively high tellurium contents.

Tellurium assaying is laborious and costly, and sensitivities lower than 100 ppm are difficult to attain. Experiments at the Bureau of Mines, Salt Lake City Research Center, resulted in development of a method for determining tellurium in amounts between 0.5 and 1.0 ppm Te.⁸ Quantitative results can be estimated by spectrographic and X-ray methods. One company has used the spectrograph to determine the tellurium content of more than 1,500 samples, having a low-sensitivity range of 10 to 100 ppm. A new sensitive test for tellurium based on the induced precipitation of elemental gold was published in 1963.⁹

The Bureau of Mines published the description of a field test for tellurium and selenium of which can be performed by nonanalytical personnel. A simple qualitative test for native tellurium and tellurium in some telluride minerals, including gold, silver, lead, and bismuth, but not including mercury or selenium, is as follows: Drop a few grains of the mineral in a test tube containing 2 or 3 cc of hot but not fuming concentrated sulfuric acid. Tellurium is indicated by the presence of a crimson-colored trail following the mineral grains.

Summary of Sample Results

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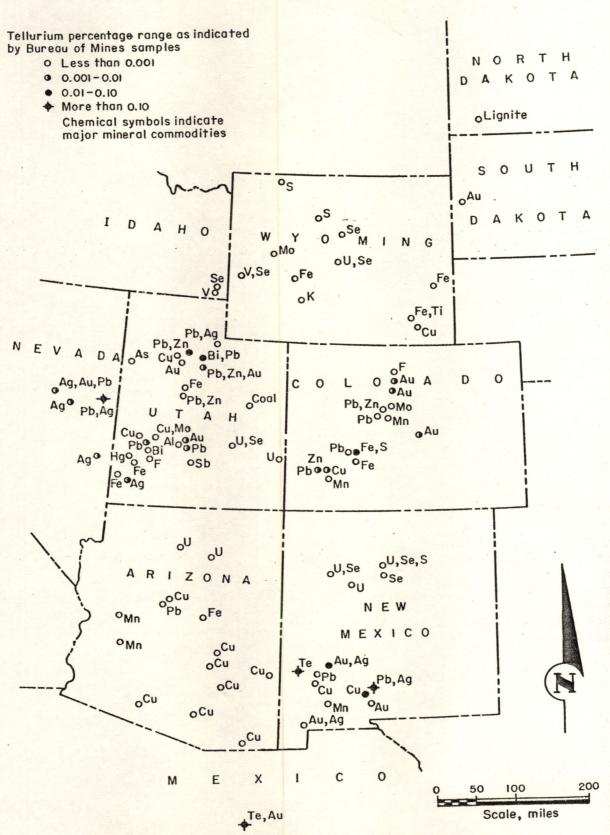
The present tellurium investigation included analysis of more than 200 samples by the Bureau of Mines (fig. 1). Approximately 62 percent of the samples contained less than 0.001 percent Te (10 ppm); 19 percent, between 0.001 and 0.01 percent; 11 percent, between 0.01 and 0.1 percent; 5 percent, between 0.1 and 1 percent; and 3 percent contained more than 1 percent Te. All samples containing more than 0.1 percent Te represent ores with native tellurium or gold, silver, bismuth, and lead tellurides. Samples containing 0.001 to 0.01 percent tellurium were obtained from deposits whose chief minerals contain lead-zinc, copper, or bismuth. Tellurium in ores from deposits of this type probably would be recovered as a byproduct of copper or lead refining. Samples containing alunite, antimony, coal and lignite, fluorite, pyrite, manganese, mercury, molybdenum, phosphate, sulfur, vanadium, titanium, or uranium, showed no significant quantity of tellurium.

Samples from the Hilltop, Memphis, Lone, Pine, Banbolla, and Silver King Properties contained more than 1 percent tellurium.

Anderson, W. L., and H. E. Peterson. Determination of Tellurium. BuMines
Rept. of Inv. 6201, 1963, 9 pp.

Lakin, H. W., and C. E. Thompson. Tellurium: A New Sensitive Test. Sci., July 5, 1963, v. 141, No. 3575, pp. 42-43.

Niebuhr, Philip E., and Allan H. Macmillan. Field Test for Tellurium and Selenium. BuMines Rept. of Inv. 6006, 1962, 6 pp.



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FIGURE 1. - Map Showing Principal Areas of Tellurium Investigations.

Description of Properties

Hilltop Mine

One sample from the Hilltop mine in the Organ Mountains, Dona Ana County, N. Mex., 11 consisted principally of galena with tellurium present as altaite, the lead telluride. Other samples from the Hilltop contained less tellurium. Sulfide minerals can be found in several working places, but generally occur in thin seams. The property is within the White Sands Military Reservation and currently is restricted from mining development.

Memphis Mine

An oxide sample from the bismuth cut at the Memphis copper mine, also in the Organ Mountains of New Mexico, assayed more than 1 percent tellurium. The mineral tetradymite had previously been reported from this location. However, additional sampling of the bismuth cut indicated that most of the zone contains less than 10 ppm tellurium.

Lone Pine Mine

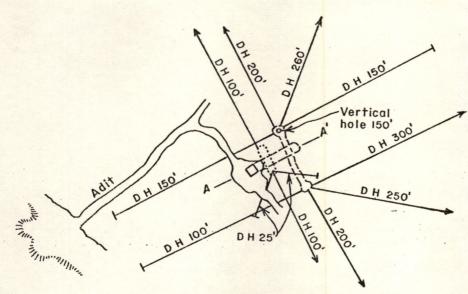
The Lone Pine mine, formerly known as the Tommy Knocker, Lausen, and Swartz, is part of a property comprising six claims in the Wilcox district, Catron County, N. Mex., which was under lease and option to the Minnesota Mining & Manufacturing Co. in 1961 and 1962. The property is 8 miles east of the intersection of Little Dry Creek and U.S. Highway 260, 13 miles south of Glenwood. Two miles of rough, winding road was constructed to the mine in 1960 over the steep front of the Mogollon Range. The property has been developed by more than 1,000 feet of workings, comprising several shallow shafts and one principal adit and shaft. The workings are in a zone of altered and pyritized Tertiary quartz latite porphyry, underlain by andesite. 13 The principal development consists of a 200-foot shaft (partly sunk as a winze) at an incline of about 80° and approximately 700 feet of adit. The adit intersects the shaft 135 feet from the portal. The shaft was sunk 200 feet below the adit level. About 20 feet of drift has been driven at the 45-foot level and about 70 feet at the 160-foot level of the shaft. The project conducted by Minnesota Mining & Manufacturing Co. consisted of road building; surface mapping; diamond drilling from the surface; shaft rehabilitation; drifting at the 45- and 160-foot levels; underground diamond drilling from the adit, 45-, and 160-foot levels; underground diamond drilling from the adit, 45-, and 160-foot levels; and sampling (fig. 2).

First work on the Lone Pine property was probably done during the early 1930's, reportedly some gold ore was shipped. Approximately 3 tons of high-grade ore containing native tellurium was sorted and piled on the dump. Most

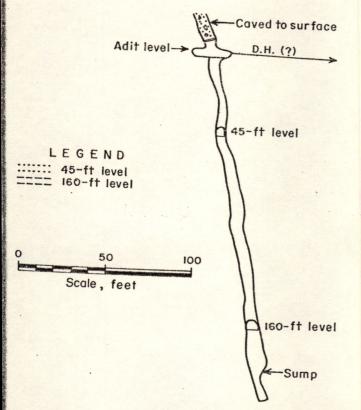
New Mexico Sch. Mines and State Bur. Mines and Miner. Res., Bull. 11, 1935, 272 pp.

¹² Work cited in footnote 10.

¹³Ferguson, H. G. Geology and Ore Deposits of the Mogollon Mining District, N. Mex. Geol. Survey Bull. 787, 1927, 100 pp.



PLAN, PART OF ADIT LEVEL



SECTION A-A, THROUGH WINZE

FIGURE 2. - Sketch of Workings Showing Several Drill Holes, Lone Pine Mine, Catron County, N. Mex.

of this ore has been carried away by specimen collectors. Reportedly, the tellurium ore was mined from a quartz-fluorite-pyrite-filled fissure 12 inches wide¹⁴ encountered in sinking the shaft.

Little evidence of a significant fissure was noted in examining the property. It appears probable that the tellurium occurs in an erratically fractured zone in pockets or filled vugs. Relatively high concentrations of pyrite were found with the tellurium, and fluorite was observed randomly both near and away from the high concentrations of tellurium. Much of the latest work consisted of core drilling to probe for new ore or a geologic structure that may indicate trends of mineralization. During 1961 approximately

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Tel Ward dis and copp its in r altaite] to quart: beds. Th sample as 100-ton s have the shipped f per ton, tellurium mineral h every 5 o metal ore process.

The pappears far production base-metal Mining Co. to 60,000 started in Mining & Started from its started in

Native Tellurium
From Northwest
of Silver City,
N. Mex. Am.
Mineralogist,
v. 17, 1932,
pp. 491-492.

2 tons of high-grade native tellurium ore was recovered by scavenging work in the shaft about 100 feet below the adit. Numerous faults were exposed in the workings. Apparently most of the faults are not related to tellurium deposition. Native tellurium is easily identified as a dark gray to silvery-white, heavy metallic mineral, usually in granular form. Pyrite, and frequently coarse, crystalline fluorite, are associated with tellurium. Two samples cut across the end of the shaft assayed 0.04 and 0.06 percent tellurium.

Bambolla Mine

Native tellurium, paratellurite, emmonsite, and mackayite occur at the Bambolla mine, 11 miles southwest of Moctezuma, Sonora, Mexico. The tellurium minerals are erratically distributed throughout seven veins ranging from 3 to 6 feet in width. The veins are in a layered, welded tuff formation of Tertiary age. Quartz is the principal gangue mineral. Rehabilitation, surface, and underground mapping and diamond drilling were done by a prominent company in 1962. Reportedly, about 10,000 tons of ore containing 30,000 pounds of tellurium was developed above the 290-foot level. The ore averages 0.15 percent Te and 0.37 ounces of gold per ton. Fifty-six samples taken from the mine dump, representing about 7,000 tons, averaged 0.107 percent Te.

Silver King Property

Tellurium has been found in the ores of the Silver King property in the Ward district, White Pine County, Nev., 16 miles south of Ely. Lead, zinc, and copper sulfides, having some silver and gold, occur in replacement deposits in relatively narrow beds of limestone. Tellurium minerals hessite and altaite have been identified in the ore. Mineralization appears to be related to quartz monzonite dikes and narrow fault fissures cutting the limestone beds. The ore is found in several contiguous beds of limestone. One mine sample assayed 3.5 percent Te, and two sample pulps, each representing about 100-ton shipments, contained 0.15 percent Te. Ores with high-silver content have the greatest percentage of tellurium. About 2,000 tons of ore recently shipped from the property averaged 30 ounces of silver and 0.05 ounce of gold per ton, 0.65 percent copper, 3.8 percent lead, and 4.8 percent zinc. tellurium content of this ore, assuming that most of the tellurium is in the mineral hessite, was estimated at 0.06 percent or 0.01 percent tellurium for every 5 ounces of silver. The ore was shipped to a company purchasing basemetal ores, and most of the tellurium was reportedly recovered in the refining

OUTLOOK FOR INCREASED PRODUCTION

The possibility of increasing tellurium production above that of 1961 appears favorable. Most of any increase probably would result from greater production and improved efficiency in recovery methods in the electrolytic base-metal refining industry. In 1961, United States Smelting, Refining and Mining Co. increased production capacity at its Indiana refinery from 17,000 to 60,000 pounds of tellurium a year. A tellurium pilot-plant project was started in 1962 by Kennecott Copper Corp. at Garfield, Utah. The Consolidated Mining & Smelting Co. of Canada may program tellurium recovery as a byproduct from its smelting and refining operations.

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t y, New properties, such as the American Smelting and Refining Company Mission (copper) project, and increased activity at mines presently yeilding complex base-metal ores will result in additional tellurium production. Some ores might be developed from mines and dumps in districts virtually abandoned; however, as no credit presently is allowed for tellurium by custom smelting companies, little increase in production can be expected from this source.

Base-metal ores from several areas were sampled by the Bureau of Mines. Samples containing tellurium were obtained from the Jerome district in Yavapai County, Ariz.; Cebolla district (Vulcan-Good Hope mines) in Gunnison County, Gold Hill district in Boulder County, and the Sneffels and Telluride districts in Ouray and San Miguel Counties, Colo.; Ward district in White Pine County, Nev.; the Organ Mountains in Dona Ana County, N. Mex.; and the Stateline district in Iron County, the Marysvale district, Piute County, the Tintic district, Juab and Utah Counties, the Cottonwood and West Mountain (Bingham) districts in Salt Lake County, and the Ophir district, Tooele County, in Utah.

Minor production may come from native tellurium occurrences under investigation in western New Mexico and central Sonora, Mexico. The extent of prospecting and developing native tellurium deposits will depend on an increase in price and demand.

Tellurium recovery from gold-telluride ores has not been economically feasible. However, if the Carlton mill of The Golden Cycle Corp. at Cripple Creek, Colo., resumed operations at a capacity of 1,000 tons of ore a day, tellurium recovery might be profitable.

Concentrating and refining tellurium are not considered difficult metallurgical processes. Improved efficiency can be expected at base-metal
smelters and refineries as demand and price for tellurium increase. Recovery
of tellurium from gold-telluride ores and ores containing native tellurium has
not been practiced commercially. Any potential producer of high-grade tellurium ores or concentrates probably could make arrangements for processing with
one of the electrolytic refining companies. United States Smelting, Refining
and Mining Co. purchases special smelter and refinery slags and residues containing tellurium and associated elements for recovery in its electrolytic
refining plant at East Chicago, Ind.

APPENDIX A. - BRIEF DESCRIPTION OF THE ROLE OF THE SEMICONDUCTOR IN THERMOELECTRICITY¹

"In a metal each atom contributes at least one electron able to move freely within the metal. In semiconductors, on the other hand, only a few atoms release such free electrons. The number of electrons available for current flow in a semiconductor is hundreds or thousands of times less than in a metal, and this accounts for the low conductivity of these substances.

"When one end of a semiconductor--or a conductor, for that matter--is hotter than the other, electrons leave the hot end more often than they do the cold end. They tend to flow toward the cold end, and since all are negatively charged, the cold end soon becomes charged negatively with respect to the hot end After a short time the flow of electrons from the hot end reaches equilibrium with the return flow from the cold end. The charges no longer accumulate, but the cold end remains negatively charged. The fewer the electrons available for the return flow, the higher will be the voltage attained at the cold end before equilibrium is reached. Since the number of free electrons is much smaller in a semiconductor, a temperature difference in a semiconductor will produce a much greater voltage than in a conductor. This is the essence of the thermoelectric advantage possessed by semiconductors.

"Different conductors develop greater or lesser voltages. In all semiconductors the voltage increases with the difference in temperature between the hot and cold ends. The voltage across a given semiconductor when one of its ends is warmer than the other is the measure of its characteristic thermoelectric power, which is expressed in volts per degree centigrade

"Semiconductors possess still another thermoelectric advantage not found in metals at all. In some types of semiconductor material the voltage differential between the hot and the cold end is set up not by the flow of negatively charged electrons but by the flow of positively charged "holes" vacated by electrons. As a result, the cold end in such a semiconductor becomes positively charged. The two types of semiconductors are designated as "n-type" (hot end positive) and "p-type" (cold end positive). In both types, of course, the direction of the current (electron flow) is from the positive to the negative end." (Fig. 3.)

An example of a simple thermoelectric generator is shown on figure 4.2

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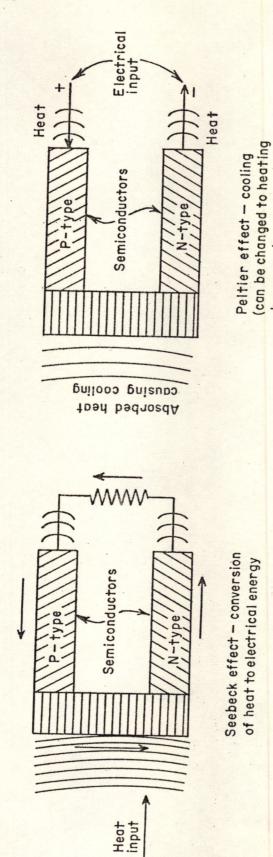
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¹ Excerpts from Joffe, Abram F. The Revival of Thermoelectricity. Sci. Am., v. 199, No. 5, Nov. 1958, pp. 31-37. (Reprinted with permission. Copyright 1958 by Scientific American, Inc. All rights reserved.)

² Abel, Edward O. Thermoelectric Power From Lead--What's Practical? What's Possible? Pres. at 34th Ann. Meeting, Lead Ind. Assoc., Inc., St. Louis, Mo. April 11, 1962.



by reversing current) FIGURE 3. - Sketch Illustrating Seebeck and Peltier Thermoelectric Effects.

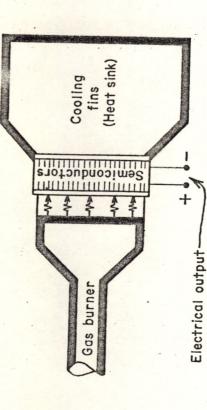


FIGURE 4. - Simple Thermoelectric Generator Housed in a Heat Exchanger.

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APPENDIX B. - PROPERTIES SAMPLED AND TELLURIUM AND SELENIUM DETERMINATIONS BY THE BUREAU OF MINES

Property	Principal		Te,1	Se,1	Remarks
	commodity	Arizona	percent	percent	
Douglas Reduction Works.	T _C	-			T-
Douglas Reduction works.	Cu	Cochise	D	-	Lavender pit ore
Do	0	1			concentrates.
Do	Cu	do	D	-	Reverberatory matte.
Do	Cu	do	E	-	Reverberatory slag.
Do	Cu	do	D	-	Cu.
Orphan Boy	U ₃ O ₈	Coconino	E	Е	Composite of 6 shipments.
Rare Metals Mill	U308	do	E	E	Composite of U ores.
Pig Iron claims	Fe	Gila	E	-	Low-grade magnetite.
Do	Fe	do	E	- /	Do.
Inspiration Consolidated Copper Co.	Cu	do	Ė	-	Cu ore concentrate.
Do	Cu	do	Е		Powerhoustons also
Do	Cu	do	E	_	Reverberatory slag.
Do	Cu	do			
Copper Cities	Cu		E E		Blister Cu.
Phelps Dodge Morenci	Cu	do			Cu ore concentrate.
		Greenlee	D	-	Flotation concentrat
Do	Cu	do	Е	•	Smelter reverberator slag.
Do	Cu	do	E	-	Cu matte.
Do	Cu	do	D	-	Anode Cu.
rtillery Peak	Mn	Mohave	E	-	Maggie Canyon ore.
Mendon Stockpile	Mn	do	E	-	Mn ore from Arizona.
anner	Cu	Pima	E	C	Cu concentrate.
helps Dodge-Ajo	Cu	do	E	-	Flotation concentrat
Do	Cu	do	E	-	Reverberatory slag.
Do	Cu	do	E	-	Cu matte.
Do	Cu	do	E	-	Anode Cu.
ohnson's Camp mine	Cu	do	E	-	Concentrate composit
an Manuel	Cu	Pinal	E	C	Cu concentrate.
Do	Cu	.do	E		MoS, concentrate.
ron King	Cu	Yavapai	E	-	Cu concentrate.
Do	Cu	do	E	C	Pyrite sample.
nit - 1 ** 1	Cu	do	E		Cu 3.08 percent 300
Do	Cu	do	D	С	level. Cu 1.05 percent 300
					level.
	Cu	do	Е	D	Cu 3.8 percent 4500 level.
Do	Cu	do	D	В	Cu 0.5 percent 700 level.
		Colorado			
rlington	CaF ₂	Boulder	E		Composite of ore.
	Au	do	c		Mineral sample.
Do	Au	do	E		Dump sample.
orman	A11	do	n		Do
= Te or Se above 1 per		40	D		Do.

A = Te or Se above 1 percent; B = Te or Se between 0.1 and 1.0 percent; C = Te or Se between 0.01 and 0.1 percent; D = Te or Se between 0.001 and 0.01 percent; and E = Te or Se of less than 0.001 percent.

APPENDIX B. - PROPERTIES SAMPLED AND TELLURIUM AND SELENIUM DETERMINATIONS BY THE BUREAU OF MINES (Con.)

Property	Principal	1	Te,1	Se,1	Remarks
	commodity			percent	
		Colorado (Con			
Cash		Boulder	В	-	Au 13.48, Ag 214.0 oz/ton.
Rex	Au,Ag	do	В	-	Au 39.22, Ag 197.7 oz/ton.
Vulcan	Au, Ag, Cu, Pb	Gunnison	D	-	S dump, Au 0.06, Ag 2.0 oz/ton.
Do	Au, Ag, Cu, Pb	do	С	-	Vulcan dump.
Do	Au, Ag, Cu, Pb	do	Е	-	Pyrite nodule.
Cebolla Creek	Fe,Ti	do	E	-	Titaniferous Fe.
Hayden shaft	Pb,Zn	Lake	Е	-	Pb 2 percent, Mn 9 percent.
Leadville	Mn	do	E	-	Mn ore.
Red Mountain mill	Pb	Ouray	D	D	Pb concentrate.
Do	Zn	do	E	D	Zn concentrate.
Do	Cu	do	C	C	Cu concentrate.
Do		do	E	D	Composite of ore.
Do		do	E	E	Mill tailing.
unnyside	Mn,SiO	San Juan	E	-	Rhodonite deposit.
andora mill	Zn	San Miguel.	E	D	Zn concentrate.
Do	Cu	do	D	D	Cu concentrate.
Do		do	E	E	Composite of ore.
Do	Pb	do	D	C	Pb concentrate.
Do		do	E	E	Mill tailing.
acky Strike	Pb,Zn	Summit	E	-	3
ilfley	Pb,Zn	do	E	-	
okomo	Pb,Zn	do	E	-	Bureau of Mines project.
limax		do	E	-	Pyrite concentrate.
he Golden Cycle Corp	Au	Teller	D	-	Au ore.
. Do	CONTRACTOR OF THE PARTY OF THE	do	C		Au flotation
					concentrate.
		Idaho			· ·
loomington Canyon	V_0_	Bear Lake	Е	D	Vanadiferous shale.
aris Canyon	N O E	do	Е	С	Metals Reserve stockpile.
		Nevada			
Consolidated Eureka	Cu,Pb,Zn, Ag	Eureka	С	-	Ore shipments to smelter.
ristol Silver	Ag	Lincoln	C	E	Do.
ilver King		White Pine.	С	D	Ore sample.
Do	Ag, Pb, Zn	do	C	C	Do.
Do	Pb	do	A	C	Galena.
Do	Ag,Pb,Zn Cu	do	С	E	Oxide ore.

¹A = Te or Se above 1 percent; B = Te or Se between 0.1 and 1.0 percent; C = Te or Se between 0.01 and 0.1 percent; D = Te or Se between 0.001 and 0.01 percent; and E = Te or Se of less than 0.001 percent.

APPENDIX B. - PROPERTIES SAMPLED AND TELLURIUM AND SELENIUM DETERMINATIONS BY THE BUREAU OF MINES (Con.)

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Property	Princip	all Count			-	
	commodi			e,1	Se	Remarks
			perc	cent	perc	ent
Silver King	Ag, Pb, Z	Nevada (C	on.)			
	10, ,-	n, White Pin	e. C		1 -	Ore bin.
Do	Cu					ore bin.
	10, -,-	n, do	B	1	-	Ore in out
Do	Cu					Ore in Quinn stope.
	Ag, Pb, Zr	1, do	C		-	One i e i
	Cu					Ore in Quinn stope
Do						20 feet higher than
Do	Ag, Pb, Zn	do	. D			above.
						Drill hole 1, 94 to
Do	Ag, Pb, Zn	do	D			96 feet.
[12] 전 경영·경영·경영·경영·경영·경영·경영·경영·경영·경영·경영·경영·경영·경		1	· D		•	Drill hole 1, 160 to
Do	. Ag, Pb, Zn	do	1 -			1/2 feet.
		40	· D		-	Drill hole 1, 194 to
Do	. Ag, Pb, Zn	1				196 feet.
	Cu	, do	· C		-	High-grade Pb-Zn,
Do	. Ag, Pb, Zn	1				Quinn stope.
	C	do	· C	1	-	Five-foot cut, Quinn
Hamilton Mining Co	AgaPh					stope.
	126 10	do	D		-	Ore shipments to
						smelter.
Lone Pine	. Te	New Mexic	0			Tome It et .
Do	· Te	Catron	A	T	C	Pyritized rock.
Do	· Te	do	В		-	Do.
Do		do	C		-	Do.
	· Te	do	В		_	
Do						Two-ton ore pile on
	Te	do	В		_	dump.
Do						Adit station chip
Do	Te	do	В			sample.
Do	Te	do	E			Do.
			1 1		-	Quartz latite porphyry
Do				1		145-foot level of
Do	Te	do	C	1		shaft.
D-			C		-	Site of shaft, 100-
Do	Te	do	Е	1		foot level.
De			E		E	Surface latite
Do	Te	do	77			porphyry.
Do	Te	do	E		E	Do.
Do	Te	do	В		E	Pyritized porphyry.
Do	Te	do	C	1	D	Do.
Do	Te	do	A	1	C	Pyrite and native Te.
Do	Te	do	В	1	C	Do.
Do	Te		C		-	Dump sample.
		do	C		-	Mud in 1/2-inch fault
Do	Te	1-				in shaft.
		do	A		-	
						Native Te and pyrite,
						Bi 0.05 percent, Au
= Te or Se above 1 pero Se between 0.01 and 0.1						4.00 oz, Ag 0.6
- above i her	onr D - m	ST THE STATE OF TH				UZI LUII.

Se between 0.01 and 0.1 percent; B = Te or Se between 0.1 and 1.0 percent; C = Te or and E = Te or Se of less than 0.001 percent.

APPENDIX B. - PROPERTIES SAMPLED AND TELLURIUM AND SELENIUM DETERMINATIONS BY THE BUREAU OF MINES (Con.)

Property	Principal		Te,1	Se,1	Remarks
	commodity		percent	percent	
Texas Main	Au, Ag, Pb	lew Mexico (-
Do	Au Ag Ph	Dona Ana.	E	-	Mine dump near crest
		do	E	-	Third mine dump from crest.
Do		do	Е	-	Fifth mine dump from crest.
Do		do	Е	-	Five-foot cut in adi of principal workings.
Do		do	D	-	One-foot cut on fis- sure in adit.
Do	Au, Ag, Pb	do	D	_	Lead ore on dump.
Do	Au, Ag, Pb	do	Е	-	Dump sample (4,000 tons).
Do		do	Е	-	Galena and sphalerite pieces.
Do	Au, Ag, Pb	do	E	-	Hematite sample.
Do	Au, Ag, Pb	do	E		Cross fissure dump.
Memphis	Au, Ag, Pb,	do	A	-	Oxidized mineral in bismuth cut.
Do	Au, Ag, Pb, Zn, Cu	do	Е	-	Large sample iron and copper oxidized
Do	Au, Ag, Pb,	do	Е	-	material. Cut sample, lower end
Do	Au, Ag, Pb, Zn, Cu	do	D	-	zinc shaft. South dump near
Do		do	Е	-	highway. Roos shaft dump.
Do		do	D	-	South end of bismuth pit.
Do		do	D	-	Bismuth ore, representative grab
ig Three	Au, Ag, Cu	Dona Ana.	D	-	sample. Pyrite and limonite
ady Hopkins	. Au Ag Cu	do	Е		ore.
illtop	Au, Ag, Pb	do	D		Dump. Claim 8, thin vein of
Do	Au, Ag, Pb	do	D	-	galena. Claim 5, thin vein of galena.
Do	Au, Ag, Pb	do	В	_	Claim 3, up raise.
Do	Au, Ag, Pb	do	C	-	Do.
Do	Au, Ag, Pb	do	c	-	Claim 1, adit, thin vein.
Do	Au, Ag, Pb	do	D	- 0	Claim 4, sulfides.
Do	Au, Ag, Pb	do	A	- 10	laim 8 galons

1A = Te or Se above 1 percent; B = Te or Se between 0.1 and 1.0 percent; C = Te or Se between 0.01 and 0.1 percent; D = Te or Se between 0.001 and 0.01 percent; and E = Te or Se of less than 0.001 percent.

APPENDIX B. - PROPERTIES SAMPLED AND TELLURIUM AND SELENIUM DETERMINATIONS BY THE BUREAU OF MINES (Con.)

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Property	Princi		County	Te,1	Se,1	Remarks
	commod		ew Mexico (C		percent	
Cactus Flats, south of	lAu	144	Grant	1 E		Towards
Dry Creek.	Au		Glanc	E	-	Quartz.
	Pb, Cu		do			
Ground Hog			do	E	-	Ore sample.
Clemney mine, Hachita	Au, Ag		Hidalgo	Е	-	Quartz vein filling.
Creeper mine, Hachita Mts.	Au,Ag		do	D	-	Ore sample.
Do	Au, Ag		do	D	-	Do.
Demming	Mn		Luna	E	-	Mn stockpile.
Phillips Petroleum Ann Lee shaft.	U ₃ O ₈		McKinley	Е	С	U ₃ 0 ₈ 0.201 percent.
Calumet and Hecla	U O		do	E	В	Ore stockpile.
Homestake Sec. 15 drill hole.	0308 0308		do	E	C	Uranium ore sample.
La Bajada	U308		Sandoval	E	-	Dump sample.
Drill hole 64	Se		do	E	-	Sandstone and silt- stone with 0.4 oz Ag and about 0.003
						percent Se.
Drill hole 81	Se		do	Е	D	Sandstone and silt- stone with 0.4 oz Ag
Drill hole 82	Se		do	E	-	Do.
Silver Tail	Au, Ag	1	Sierra	C	_	Small quartz fissure.
			North Dakot			
Lignite beds	U, O,	T	Bowman	E	C	
Photography and the second second			South Dakot	a		
Homestake	Au	1	Lawrence	Е	Е	Minus 3/4-inch mill head.
Do	Au		do	E	D	Slag from cyanide
Do	Au		do	E	D	precipitate. Spiral concentrator
Do	Au		do	E	E	sulfide concentrate. Sand charge to spiral
Do	Δ11	-	do	E	E	concentrator. Slime fraction.
D-	Au	8	do			
				Е	E	Iron-stained schist ore.
14.	Au		do	E	E	Schist rock cut near Yates shaft.
			Utah			lates shart.
Creole	Ag, Pb	Ti	Beaver	E	D	Ore sample.
	Ag, Pb	1	do	E		Bi 0.3 percent.
'not-	Cu		do	E		
lk mi-	Cu		do	E	E	Copper ore.
'Opp Tr'	Cu					Do.
			do	Е		Chalcopyrite in dump ore.
A = Te or Se above 1 per	cent: R	= 7	Te or Se hetr	reen 0 1	and 1 0	percent: C = To or
Se between 0.01 and 0.	1 perce	nt.	D = Te or S	hetroer	0 001	and 0 01 persont.
and E = Te or Se of le	on than	0,0	01	DOCWCCI	. 0.001	and o.or percent,

and E = Te or Se of less than 0.001 percent.

APPENDIX B. - PROPERTIES SAMPLED AND TELLURIUM AND SELENIUM DETERMINATIONS BY THE BUREAU OF MINES (Con.)

Property	Principal	County	Te,1	Se,1	Remarks
	commodity		percent	percent	
		Utah (Con.)			
Moscow	Zn	Beaver	E	-	Zinc ore.
Horn Silver	Au, Ag, Pb,	do	D	-	Ore sample.
	Zn				
Staats Fluorite	CaF	do	E	E	U ₃ 0 ₈ and CaF ₂ in ore.
Do	CaF	do	Е	E	Composite CaF ₂ .
Utah Western Antimony	Sb	Garfield	E	E	Ore grab.
AEC No. 8	U308	Grand	E	В	U ₃ 0 ₈ 0.327 percent.
Do	U308	do	E	E	Mainly pyrite.
Cina	Hg	Iron	E	E	Pit sample.
Do	Hg	do	E	E	Hg ore.
Iron Mountain	Fe	do	E	Е	Iron ore.
Chief Consolidated	Pb,Zn	Juab	E	-	Fe 42.8 percent.
Deer Trail	Au, Pb, Zn	Piute	D	C	Tailings.
Do	Au, Pb, Zn	do	D.	C	Dump sample.
Lucky Boy	Au	do	D	A	Sample of tiemannite
					(HgSe).
Goldstrike	Au	do	D	D	Grab of ore.
Cascade	Au	do	D	D	Do.
Shamrock	Au	do	D	D	Do.
Bully Boy	Au	do	E	D	Do.
Do	Au	do	D	E	Mill tailing.
Iris	Au	do	D	D	Grab of dump.
Great Western	Au	do	D	D	Grab of ore.
Packard	Au	do	E	E	Dump.
Tate	Au	do	E	E	Composite of 4 cuts.
Blue Bird	Au	do	E	E	Dump grab.
Kimberly	Au	do	E	E	Do.
Do	Au	do	E	E	Ore-bin grab.
Do	Au	do	E	E	Tailing pond.
Annie Laurie	Au	do	E	E	Do.
Do	Au	do	E	E	Ore-bin grab.
Winkleman	Alunite	do	E	-	Pit grab.
White Horse	Alunite	do	E	-	Do.
South Hecla	Pb,Zn,Au	Salt Lake	D	-	Ore sample. Bi winze ore.
Do	Pb,Bi	do	C	_	Cu ore concentrate.
Utah Copper mine	Cu	do	D E		Reverberatory slag.
Do	Cu	do			Cu matte.
Do	Cu	do	D	_	Converter Cottrell
Do	Cu	do			dust.
		1.	D		Anode Cu.
Do	Cu	do	D	D	Pb concentrate.
U.S. and Lark mine	Cu, Pb, Zn,	do	D	D	To concentrate.
	Ag	do	C	P	Pb bullion.
Do	Pb	do	C E	E	Ore sample.
Keetley unit, United	Pb,Zn	Summit	L	E	ore sampre.
Park City Mines. 1 A = Te or Se above 1 pe	I P -	To or Se het	ween 0	1 and 1.	O percent: C = Te or

1A = Te or Se above 1 percent; B = Te or Se between 0.1 and 1.0 percent; C = Te or Se between 0.01 and 0.1 percent; D = Te or Se between 0.001 and 0.01 percent; and E = Te or Se of less than 0.001 percent.

APPENDIX B. - PROPERTIES SAMPLED AND TELLURIUM AND SELENIUM DETERMINATIONS BY THE BUREAU OF MINES (Con.)

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Property	Princ	ipa	1 County	Te,1	Se,1	Remarks
	commo				percen	
220			Utah (Con.)	The second	
Mayflower New Park		1	Summit	E	E	Ore sample.
Sacramento pit			Tooele	E	E	Black shale.
Consolidated Mercur			do		E	Au ore tailings.
Do	Au		do		E	Do.
Manning Canyon	Au		do		-	Do.
Mercur Dome			do			Au ore.
Ophir Hill			do		D	
				1 "	1 "	Shale between Blue
Do	Pb.Zn		do	D	D	and Green veins.
Do			do		D	W stope.
	,		40	1	D	Grab from Pb-Zn stope,
Do	Ph Zn		do	1 -		1600 level.
	10,211		40	D	D	Cu ore, Green vein,
Gold Hill mine	As		100	-		1400 level.
U.S. mine	As		do	E	-	Arsenopyrite ore.
International Smelting	Pb		do	E	-	Do.
and Refining Co.,	10		do	E	D	Pb blast-furnace
Tooele plant.						slag.
Do	DL.					
Do	Pb		do	В	В	Cottrell dust.
DO	Pb		do	В	В	Blast furnace bag-
D-						house fumes.
Do	Cd		do	C	В	Cd plant calcine.
Do	Cu		do	C	C	Dross reverberatory
** AND 1719						matte.
Do	Cu		do	C	C	Dross reverberatory
						speiss.
	РЬ		do	C	E	Pb bullion.
	Fe		Utah	E	E	Coal to furnace.
Plant.				TO THE		
Do	Fe		do	E	-	Blast-furnace flue
						dust.
	Fe	1	do	E	-	Pig iron.
Do	Fe	.	do	E		Open-hearth flue dust.
Silver Reef	Au, U, O	.	Washington.	. В	-	Au ore sample.
	0 0		Wyoming			The state of the s
Iron Mountain	Fe.Ti		Albany	E		Tit if
Lysite Poison Butte	Se		Fremont	E	-	Titaniferous magnetite.
			L'emoire	L.	C	Sample contains 0.049
Lucky Mc	J ₃ 0 ₈		do	P		percent Se.
	3 8		do	E	В	U ₃ 0 ₈ 0.023 percent,
Do	10		do	F		Se 0.825 percent.
	J ₃ 0 ₈		do	E	C	Unoxidized ore, U ₃ 0 ₈
Frazier-Lamac	10		do	P		0.45 percent.
Vitro Sateco mine	J 0 8		do	E		Ore with pyrite.
Atlantic City iron mine.	3 8		do	E	C	Do.
1 A = Te or Se above 1		1	do	E	- 1:	Taconite ore.
A = Te or Se above 1 perc	ent; B	= '	te or Se betw	reen 0.1	and 1.0	percent; C = Te or

Se between 0.01 and 0.1 percent; D = Te or Se between 0.1 and 1.0 percent; C = Te or Se between 0.01 and 0.1 percent; D = Te or Se between 0.001 and 0.01 percent; and E = Te or Se of less than 0.001 percent.

APPENDIX B. - PROPERTIES SAMPLED AND TELLURIUM AND SELENIUM DETERMINATIONS BY THE BUREAU OF MINES (Con.)

Property	Principal commodity		Te,1 percent	Se,1 percent	Remarks
		Wyoming (Cor		**	
Brutch Sulfur	S	Hot Springs		В	Sulfur ore.
Copper King	Cu, Au	Laramie	E		Drill-hole sample.
Dry Canyon	V_0_	Lincoln	E	C	Vanadiferous shale.
Do	v 0	do	E	C	Do.
Sunrise	Fe0	Platte	E	-	Hematite ore.
Sunlight Basin	S	Park	E	-	S ore.
Fremont Lake	MoS	Sublette	E	-	Cut sample.
Wyomingite rock	K ₂ 0°	Sweetwater.	E	-	Leucite mineral high in potash.

¹A = Te or Se above 1 percent; B = Te or Se between 0.1 and 1.0 percent; C = Te or Se between 0.01 and 0.1 percent; D = Te or Se between 0.001 and 0.01 percent; and E = Te or Se of less than 0.001 percent.

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APPENDIX C. - SOME SPECTROGRAPHIC ANALYSES FOR TELLURIUM FROM AN INVESTIGATION BY A PRIVATE COMPANY

(Mine Locations are Listed According to State and County)

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Type Sample
Calaveras County:
California California Carson Hill Tailings Au 8 ND
Calaveras County: Carson Hill. Tailings
Calaveras County: Carson Hill. Tailings
Carson Hill
Melones
Ford No. 1. Dumps and float. Au
Ford No. 2. Dump. Au 4 ND
Melones-Morgan Float Au 9 ND Calaveras vein do. Au 1 ND Stanislaus. Outerop. Au 1 ND Keystone. Ore bin and dump. Au 3 ND Union. Dump and float. Au 2 ND Empire. Outcrop. Au 1 ND Eldorado County: Dump and float. Au 2 ND Sheridan. Dump. - 7 ND Fresno County: Outcrop. - 1 ND Sheridan. Outcrop. - 1 ND Garnet Dike. Dump, tailing, and underground samples. ND ND Inyo County: Susquehanna. Outcrop. - 1 L Susquehanna. Outcrop. - 1 L Do. do. - 1 ND Sampson-Eila Underground and dump - 4 ND
Calaveras vein
Stanislaus
No
Union. Dump and float. Au 1 ND
Empire. Outcrop. Au 1 ND Eldorado County: Darling. Dump 7 ND Fresno County: Sheridan. Outcrop 1 ND Dinky Creek. Ore sacks. WO3 1 L Garnet Dike. Dump, tailing, and underground samples. Inyo County: Susquehanna. Outcrop 1 L Do. do 5 ND Sampson-Ella. Underground and dump 4 ND Cerro Gordo. Underground. Pb 1 ND Concentrate Pb 1 L Do. Concentrate Pb 1 L Do. Dumps. Pb 16 ND Coso Hot Springs. Outcrop 2 ND Devil's Kitchen. do 5 ND Big Horn. Underground 1 ND Limperial County: - 4 ND Limperial County: - 4 ND Outcrop 2 ND Limperial County: - 4 ND Outcrop 4 ND
Darling. Dump. - 7 ND
Darling Dump.
Presno County: Outcrop. Out
Sheridan
Dinky Creek
Dump, tailing, and underground samples.
Inyo County: Susquehanna. Outcrop. - 1 L L Outcrop. - 5 ND Outcrop. - 1 ND Outcrop. - 2 ND Outcrop. - 2 ND Outcrop. - 1
Inyo County: Susquehanna. Outcrop. - 1 L County: Susquehanna. Outcrop. - 5 ND ND Coso Hot Springs. Outcrop. - 1 L Coyote Mountains. Outcrop. - 1 L ND Coto
Susquehanna
Do.
Bendiro Iron.
Sampson-Ella. Underground and dump 4 ND do 1 L Cerro Gordo. Underground. Pb 1 ND Do. Concentrate. Pb 1 L Do. Dumps. Pb 16 ND Devil's Kitchen. do 2 ND Golden Treasure. Dump 1 ND Big Horn. Underground 1 ND Imperial County: - 4 ND Outcrop 4 ND
Do.
Cerro Gordo. Underground. Pb 1
Do Concentrate. Pb 1 1 L ND Do Dumps. Dumps. 2 ND Outcrop. 2 ND ND Outcrop. 3 ND ND Devil's Kitchen. Dump. 3 ND Dump. 3 ND ND ND Dump. 3 ND
Do. Do. Do. Do. Coso Hot Springs. Devil's Kitchen. Golden Treasure. Dump. Big Horn. Underground. Dump. Underground. Dump. Underground. Dump. Dum
Coso Hot Springs Devil's Kitchen Golden Treasure Big Horn Underground Coyote Mountains Outcrop 2 ND
Devil's Kitchen Golden Treasure Big Horn Underground Coyote Mountains Dump Underground Outcrop ND ND ND ND ND ND ND ND ND
Golden Treasure Big Horn Underground Imperial County: Coyote Mountains Outcrop In ND ND ND ND
Big Horn Underground 1 ND Imperial County: - 4 ND Coyote Mountains Outcrop 4
Imperial County: Coyote Mountains Outcrop 4 ND
Coyote Mountains Outcrop
IDS Ancolog County
Los Angeles County: Pulp 1 ND
Mono County:
Comanche Dumps 2 ND
Nevada County:
Providence do
Plumas County:
ThunderbirdOutcrop and stockpile 15 ND
Butto Por Outcrop 4 ND
DoStockpile
Little Nell - Outcrop
S = 0.1 to 1.0 percent; M = 0.01 to 0.1 percent; L = 0.001 to 0.01 percent; and

ND = none detected.

ND = none detected.

APPENDIX C. - SOME SPECTROGRAPHIC ANALYSES FOR TELLURIUM FROM AN INVESTIGATION BY A PRIVATE COMPANY (Con.)

(Mine Locations are Listed According to State and County)

Property	Type sample	Principal elements	Number of samples	Percentage Te, spec- trographic analysis
	California (Con.)			
Riverside County: Calzona Lost Horse	Outcrop and dump Dump, tailing, and outcrop.	-	8	ND ND
San Bernardino County: Stateline	Outcrop do Sorted ore Dump do Ore bin.	- Au - - -	2 3 2 2 3 2	ND ND L ND ND
Shasta County: Deep Pit Wallis Yankee John Do Do	Outcrop do do Concentrate		2 4 1 2 2 2	ND ND ND L M
Eureka Sierra County: Plumbago Siskiyou County:	Outcrop	-	2 2 2	ND ND
Yellow Rose Sonoma County: Geyers Mercury mine	Outcrop	- Hg	2 1	ND ND
Trinity County: Golden Jubilee Dorleska		-	15	L ND
Tulare County: Beasley claims	. Dump and outcrop	-	9	ND
Tuolumne County: Jumper mine Ashley Atlas Chilana Jackass Hill Eagle Shawmut	Dumps and underground. Outcrops and dump Underground and outcrop	- - - - Cu,Zn	4 4 8 7 2 11	ND ND ND ND ND
Norwegian	tailing. Dumps and outcrop	L = 0.001 to	7 1 1 0.01 perc	ND ND ND

APPENDIX C. - SOME SPECTROGRAPHIC ANALYSES FOR TELLURIUM FROM AN INVESTIGATION BY A PRIVATE COMPANY (Con.) (Mine Locations are Listed According to State and County)

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Property	Type sample	Principal elements	Number of samples	Percentag Te, spec- trographic analysis1
	California (Con.)			did Ly 513-
Yuba County: Liberty-Red Cross Red Raven	Float and dump	-	1 2	ND ND
P- 11 0 1	Colorado			
Clyde. Wolf Tongue. Mountain Top. Do. White Raven. Do. Rex.	Ore Underground sample Underground do Ore do do do Mine tailing Dump. Mill tailing Underground Ore (6 tons) Ore (3 tons) Ore (7 tons) do Ore	CaF ₂ , Au Au CaF ₂ Au	1 3 1 1 1 1 1 1 1 1 3 1 1 1 1 1 1 1	L ND ND L ND
Nevo	do	Au	1	L
Sedalia. Stonewall. Mary Murphy. Do. Banker. Lilly. Clear Creek County:	dododododill tailingJndergrounddododododododo.	Cu, Pb, Ag Cu, Pb, Zn Pb, Zn Pb, Zn Pb, Zn Pb, Zn Pb, Ag	2 1 1 1 1 1	ND ND ND ND ND
Terrible Dunderberg. Central. Empire. Gem. Co. I	Ore	Ag,Au Ag,Au Ag,Au Ag,Au Ag,Au Ag,Au Ag,Au Ag,Au	5 1 1 1 1 1	ND ND ND ND ND ND
Forge Hill	do	Ag,Au Ag,Au = 0.001 to 0.	1 1 01 percent	L ND: and

APPENDIX C. - SOME SPECTROGRAPHIC ANALYSES FOR TELLURIUM FROM AN INVESTIGATION BY A PRIVATE COMPANY (Con.)

(Mine Locations are Listed According to State and County)

Property	Type sample	Principal elements	Number of samples	Percentag Te, spec- trographi analysis
Clear Create Co.	Colorado (Con.)			dialys15-
Clear Creek County (Con.): Franklin-Silver Age	Ore	Ag, Au	1	
Do	do	Ag, Au	2	ND
Do	do	Ag, Au	2 2	L
Custer County:		Ag,Au	2	M
Bassick	Tailing	Au	1.	
Bull Domingo	Ore	Au, Ag, Pb		· ND
Silver Cliff	do	Au, Ag	1 1	ND
Eagle County:		,8	2	ND
New Jersey Zinc	Pb concentrate	Pb		
Li rase county:		FD	6	L
The Golden Cycle Corp.	Mill tailing	1		
(at Colorado Springs).		Au	2	ND
Gilpin County:				
Miscellaneous	Mine samples			
377	Ore shipments	Au	7	ND
Gunnison County:	ore burpments	Au	4	ND
Title T	Ore			
Sylvanite	do		1	S
Vulcan	do	Au	4	ND
Do	do	Au, Ag, Cu, Pb	7	ND
Do	do	Au, Ag, Cu, Pb	1	M
Ruby	do	Au, Ag, Cu, Pb	1	S
Akron-White Pine	do	Au	2.	ND
Crested Butte	do	Pb,Ag	2	ND
insdale County:	do	Pb,Ag	1	ND
Ute-Utlay	do			
Golden Fleece	do	Au, Pb	1	ND
Do	do	Au, Pb	2	ND
ake County:	do	Au, Pb	1	M
	611 to 111			
	ill tailing	-	1	ND
0	re	-	1	ND
	do	-	1	ND
	do	-	1	ND
	• • • • • • • • • • • • • • • • • • • •	Pb,Ag	1	ND
77-1	•••••••	Pb,Ag	1	ND
Platta County:	• • • • • • • • • • • • • • • • • • • •	Pb,Ag	1	ND
21. 2111				
	re	Au, Ag, Cu, Pb	1	L
Do		Au, Ag, Cu, Pb	2	ND
May Day	do	Au, Ag, Cu, Pb	1	ND
Bessie G	10	Au, Ag, Cu, Pb	4	ND
Gold King	10	Au, Ag, Cu, Pb	3	ND
Copper Hill	io	Au, Ag, Cu, Pb	1	ND
Zodomak Hill			1	
- 0.1 to 1.0 percent: M =	0.01 to 0.1 percent. I	- 0 001 +- 0 0:		

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APPENDIX C. - SOME SPECTROGRAPHIC ANALYSES FOR TELLURIUM FROM AN INVESTIGATION BY A PRIVATE COMPANY (Con.)

(Mine Locations are Listed According to State and County)

Number Percentage Te, spec-Principal of Property Type samples trographic elements sample analysis1 Colorado (Con.) Mineral County: Au, Ag, Pb, Zn 6 ND Empress..... Ore..... Au, Ag, Pb, Zn 1 L do..... Do..... 1 ND Tailing..... Au, Ag, Pb, Zn Do..... 1 ND Au, Ag, Pb, Zn Dump..... Amethyst..... 2 ND Au, Ag, Pb, Zn Ore..... Sublet Mining Co..... Au, Ag, Pb, Zn 1 L Mine dump..... Mexico..... Montezuma County: ND 1 Ore..... Pb, Zn, Au, Ag Doyle..... Pb, Zn, Au, Ag 1 ND do...... Minerals Consolidated ... Pb, Zn, Au, Ag 1 L do..... Do..... Ouray County: ND 1 Pb, Zn, Au, Ag do..... Midnight..... 2 ND Pb, Zn, Au, Ag do..... Genessee..... 3 T. Pb, Zn, Au, Ag Camp Bird..... Concentrate...... 1 ND Do..... do..... Pb, Zn, Au, Ag 1 T. Pb, Zn, Au, Ag Hough..... Park County: ND 1 Pb, Zn, Au, Ag Hall Valley prospect 1 ND Pb, Zn, Au, Ag Ore..... Buckskin Joe 1 ND Pb, Zn, Au, Ag do..... Hock Hocking 1 ND Pb, Zn, Au, Ag Hilltop..... 1 ND Pb, Zn, Au, Ag W. G. Mines, Inc..... Saguache County: 3 ND Au, Ag Rowley..... Ore..... 2 ND do..... Au, Ag St. Louis..... 1 ND Empress Josephine..... do..... Au, Ag 1 L do..... Au, Ag Do..... 1 ND Au, Ag Bonanza..... do..... Pb, Zn 1 L Concentrate..... National Mines..... Pb, Zn 1 ND (10 tons)...... Johnson Mining Co..... San Juan County: L 1 Pb, Zn, Cu, Au, Ag Lead Carbonate..... 1 ND Pb, Zn, Cu, Au, Ag do..... Lucy..... ND 4 do...... Pb, Zn, Cu, Au, Ag Lost Hope..... 1 ND Pb, Zn, Cu, Au, Ag do...... Little Dora..... ND 1 Pb, Zn, Cu, Au, Ag do..... Lettie..... Pb, Zn, Cu, Au, Ag 1 M Shenandoah Dives..... Zn concentrate..... 1 M Pb, Zn, Cu, Au, Ag Do..... Pb concentrate..... 1 ND Gold Prince.... Ore..... Pb, Zn, Cu, Au, Ag 1 ND do..... Pb, Zn, Cu, Au, Ag Engineer Mountain..... M Pb 1 Pb concentrate..... Idarado Mining Co..... Cd 1 M Cd concentrate..... Do..... 5 Do..... Cu concentrate..... Cu 1S = 0.1 to 1.0 percent; M = 0.01 to 0.1 percent; L = 0.001 to 0.01 percent; and

ND = none detected.

APPENDIX C. - SOME SPECTROGRAPHIC ANALYSES FOR TELLURIUM FROM AN INVESTIGATION BY A PRIVATE COMPANY (Con.) (Mine Locations are Listed According to State and County)

Colorado (Con.) Summit County: Wellington. Country Boy. Lucky Strike. Monte Cristo. Chautauqua. Colorado (Con.) do do do do do do	Pb,Au,Ag Pb,Au,Ag Pb,Zn	1	analysis1
Summit County: Wellington Country Boy Lucky Strike Monte Cristo Ore do do do	Pb,Au,Ag	1	1
WellingtonOreOre Country Boydo Lucky Strikedo Monte Cristodo	Pb,Au,Ag	1	
Country Boy do Lucky Strike do Monte Cristo do	Pb,Au,Ag		ND
Lucky Strike do do		1	ND
Monte Cristo do	10.411	1	ND
	Pb,Zn	3	ND
	Pb,Zn	1	ND
Teller County:	, , , , , , , , , , , , , , , , , , , ,		
Carlton mill Tailing	Au	1	ND
Mabel M Dump	Au	ī	ND
Grass Creek Tailing	Au	î.	ND
School Section mine do	Au	î	ND
EmpireDump	Au	3	ND
Vindicator do	Au	1	ND
Victor Mill tailing	Au	î	ND
	Au	2	ND
		2	ND
Orphan May Ore bin and dump	Au		ND
Nevada	1		
Esmeralda County:		(5)	MD
Goldfield	Au		ND
Goldfield Consolidated Dump, 1 million tons		(s)	L
Lander County:		(0)	***
Battle Mountain Crude ore	Cu	(s)	ND
Lincoln County:			
Delamar Ore bin and tailing dump.	Au	2	L
Bristol Silver Crude ore	Pb,Zn,Ag	(s)	L
Utah			
Beaver County:			
Creole Underground sample	Pb,Zn,Bi	3	· ND
Sulphurdale Sulfur ore	S	2	ND
Sheep Rock mine Dump	Au, Ag	1	ND
Rob Roy do	Au, Ag	1	ND
Carbonate	Pb,Zn	1	ND
Golden Reef	Pb,Zn	1	M
Karla mine Rock	Pb,Zn	1	L
Harrington Hickory	Pb,Zn	1	ND
Inspiration	Pb,Zn	1	ND
Duchesne County:			
Myton Blackbird mine Outcrop	Mn	1	ND
Murraydo	Mn	1	ND
Emery County:			
Pike-Globe mine Ore	Cu	1	ND
S = 0.1 to 1.0 percent; M = 0.01 to 0.1 percent; L		-	

ND = none detected.

² Several samples.

APPENDIX C. - SOME SPECTROGRAPHIC ANALYSES FOR TELLURIUM FROM AN INVESTIGATION BY A PRIVATE COMPANY (Con.) (Mine Locations are Listed According to State and County)

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Property	Type sample	Principal elements	Number of samples	Percentage Te, spec- trographic analysis1
	Utah (Con.)			
Grand County: Lewis & Walling property Climax Uranium property		ບູ0 ບູ30 ບູ30	1 1	ND ND
Iron County: Stateline district: Ophir mine Johnny	Dumpdo	. Au Au	1 1	M ND
Gold Spring district: Talisman Snowflake	do	Au Au Au	1 1	ND ND L
Jumbo Escalante district: Holt mine	do	Au	2	ND
Juab County: Bullion Beck Grand Central Mammoth	Ore	Pb,Zn,Ag Pb,Zn,Ag Au	1 1 2	ND ND ND
Colorado Consolidated Swansea Knight smelter	Crude oredoDump	Pb,Zn,Ag Pb,Zn,Ag	1 1 1	L ND L
Piute County: Yellow Jacket	Alunite	A1,K Au,Ag,Pb,Zn,Cu	1 1	ND S
Shamrock	Dump ore do	Au, Ag, Pb, Zn, Cu Au, Ag, Pb, Zn, Cu Au, Pb	1	L ND ND
Bully Boy Webster Iris	dodododododododo	Au, Pb Au, Pb Au, Pb	1 1 1	ND L M
Lucky Boy Dalton Deer Horn	Ore specimen	Au, Pb Au, Pb Pb, Zn, Au	1 1 1	S ND ND
Documents Documents Salt Lake County:	do	Pb,Zn,Au Pb,Zn	1	L ND
Scott mine, Big Cottonwood. Cardiff	do	Pb,Zn	1 1	ND ND
Highland Boy U.S. and Lark mine Do	Ore shipment Ore Pyrite	Pb,Zn Pb,Zn Pb,Zn	1 1	ND ND
Do Do Utah Copper mine	Pb concentrate Zn concentrate Cu ore sample	- Cu	1 1 1	L L ND
Do	Cu oxide precipitate	L = 0.001 to 0.	01 perce	nt; and

ND = none detected.

APPENDIX C. - SOME SPECTROGRAPHIC ANALYSES FOR TELLURIUM FROM AN INVESTIGATION BY A PRIVATE COMPANY (Con.)

(Mine Locations are Listed According to State and County)

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Property	Type sample	Principal elements	Number of samples	Percentage Te, spec- trographic analysis1
	Utah (Con.)			
Salt Lake County (Con.):				
Utah Copper Mine	Tailing	Cu	1	ND .
Do	Cu sulfide concentrate.	Cu	1	ND
San Juan County:				
Happy Jack	Copper concentrate	U ₃ 0 ₈	1	ND
Summit County:		3 0		
Daly West	Sulfide ore	Pb,Zn,Ag	2	ND
Gilmore	Tailing	Pb,Zn,Ag	1	ND
Atkinson	do	Pb,Zn,Ag	1	ND .
Daly No. 2	Ore	Pb,Zn,Ag	1	ND
Ontario	do	Pb,Zn,Ag	2	ND
Pacific Bridge mill	Tailing	Pb,Zn,Ag	11	ND
Tooele County:				
Gold Hill district,	Outcrop	-	1	ND
Western Utah Copper			The state of the s	
mine.				
Do	Cu oxide	-	1	ND
Do	Fe oxide	-	1	ND
Do	Underground	-	1	ND
Do	AsS, ore	-	1	ND
Kerry Dee	2	-	1	ND
Frisbee prospect		-	1	ND
Argent		Pb,Zn	1	ND
Wintah County:				
Dyer mine, Brush Creek		-	1	ND
area.		***		
Utah County:		•	2	
Eureka Standard	Sulfide dump sample	Au, Pb, Ag	1	ND
Iron Blossom No. 3	Dump	Pb, Zn, Cu, Ag	1	ND
Iron Blossom No. 1	do	Pb, Zn, Cu, Ag	-	ND
Do	Ore shipment	Pb, Zn, Cu, Ag	-	L
Apex Standard	Dump and ore	Pb, Zn, Cu, Ag	2	ND
Yankee Consolidated	Dump	Pb, Zn, Cu, Ag	•	ND
Do	do	Pb, Zn, Cu, Ag	-	L
Eureka Lilly	Ore	Pb, Zn, Cu, Ag	-	ND
Mountain View	do	Pb, Zn, Cu, Ag	-	ND
Tintic Standard	do	Pb,Zn,Cu,Ag	-	ND
Washington County:				
Silver Reef	Sulfide concentrate	Au,U308	-	ND
	1 - 0 01 to 0 1 percent: I		01 porce	nt: and

IS = 0.1 to 1.0 percent; M = 0.01 to 0.1 percent; L = 0.001 to 0.01 percent; and ND = none detected.

APPENDIX D. - PUBLISHED REFERENCES OF TELLURIUM MINERALS WITHIN THE AREA COVERED BY THIS REPORT

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		IORI			
Property or district	County				
Arizo		Tellurium minerals			
Tomostone district (14)1		Notice M.			
bisbee waiten district (13)					
imbodyampa distilli Montgomorry mine (12)					
Groom Creek district (14)	do	,			
0-1-	1	. Native Te.			
Magnolia district: Keystone, Eclipse, Logar	n P = 11				
and Kekionga mines (19).	Boulder	 Sylvanite, calaverite, hessite, petzite, 			
Gold Hill district:		coloradoite, altaite.			
Red Cloud mine (12 14)					
Red Cloud mine (<u>13</u> , <u>14</u>)		, colladymile,			
Uncle Sam mine (12)	do	,,			
Buena mine (12)		Native Tw.			
Jamestown district:	do	Native Te.			
John Jay mine (13, 14)					
Golden Age (14)	do				
Sentinel (14).					
Tierney (14)	do				
Tierney (14)	do	Do.			
Smuggler (6) Eldora district: Enterprise (10)	do	Do.			
Eldora district: Enterprise mine (12)	do	Sylvanite, petzite,			
Rico district. Johnson B. 11		other tellurides.			
Rico district: Johnny Bull mine (17)		Tellurides.			
Whitehaven district (13)	· Fremont	Do.			
Gregory mino (12)					
Gregory mine (12)	. Gilpin	Do.			
Jewelry Shop (12)	. do	Krennerite, petzite,			
		sylvanite, altaite.			
War Dance (12)	. do	Coloradoite.			
Lake City district:					
Golden Fleece mine (7)	. Hinsdale	Petzite.			
ourse wille (9)	1	Native Te.			
varcan mine (9)	. do	Weissite, rickardite.			
Lagle and Louisville	Lake	Tellurides.			
		Total fues.			
La Plate district: Gold King, Lucky Discover Bessie G, Neglected, Idaho, Durango Girl,	, La Plata	Sylvanite, calaverite.			
and May Day mines (5).					
Ouray district:					
Jonathan mine (6)	Ouran	77			
The real relief (b)		Hessite.			
Bonanza district:		Tellurides.			
Empress Josephine mine (1)	Caguacha				
	baguache	Native Te, rickardite,			
		sylvanite, krennerite,			
		empressite, hessite,			
Copper Gulch (1)		petzite, altaite.			
Underlined numbers in parentheses refer to	do	Tellurides.			
Underlined numbers in parentheses refer to items in the list of references at the					
L L vivo Late					

APPENDIX D. - PUBLISHED REFERENCES OF TELLURIUM MINERALS WITHIN THE AREA COVERED BY THIS REPORT (Con.)

Property or district	County	Tellurium minerals
Colorado (Co	n.)	
Silverton district: Barstow mine (16) Camp Bird (16) Silver Ledge mine (16) Magnet mine (16) Bear Creek district (15) Breckenridge district: Ontario mine (14) Cripple Creek district: Gold King, Mable M, Little May, Independence, Portland, Moon- Anchor, Anchoria-Island, Elkton, Bluebird, West Point, Moose, Deadwood, and Gold Sovereign (10, 11).	San Juan do do do Teller	Tellurides. Do. Hessite. Do. Petzite. Bismuth, tellurides. Calaverite, sylvanite, krennerite, petzite.
New Mexic		
Ute Creek district: Aztec mine (3) Organ Mountains district: Ben Nevis, Crested Butte, Texas Main, Memphis, Hilltop, Little Buck, Gonzales Eureka, and Rickardite	Colfax Dona Ana	Native Fe, tetradymite. Altaite, tetradymite, hessite, petzite.
mines (3, 4). Little Burro Mountains (3) Sylvanite district: Hand Car, Golden Eagle, Gold Hill, Ridgewood, Little Mildred, and	Grant Hidalgo	Tetradymite. Tetradymite, hessite.
Pearl claims (3). Red River district: Sampson, Memphis, and Independence mines (3).	Taos	Petzite.
South Dake		1 1 1 1
Blacktail district: Double Standard mine (8). Ragged Top district: Dacy Group (8) Lead-Deadwood district: Homestake mines (18).	do	Tellurides. Do. Do.
Utah		
Stateline district (2)	Piute	Tellurides. Do.
Dry Canyon district: Kearsarge mine (14) Mercur district: Golden Gate (2)	Tooele	Hessite. Tellurides.

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dymite. mite,

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