4750 0023

ABSTRACT

Geology and ore deposits of the Taylor Mining District, White Pine.

The Taylor mining district is located on the west flank of the Schell Greek Range, 15 miles south of Ely, Nevada. Rocks in the range consist of easterly-dipping Paleozoic limestone, dolomite, and shale, which have been intruded by dikes and sills of mid-Tertiary rhyolite. Silver bearing formations in the district consist of the Devonian Guilmette Limestone and its transition with the overlying Pilot Shale; and the Mississippian Joana Limestone. Known commercial deposits in the district are restricted to the upper 400 feet of the Guilmette.

The most prominent structural feature in the district is a northnorthwest trending assymetrical anticline, having a vertical west limb and a gently dipping east limb. Two prominent fracture systems are associated with this anticline, ore striking north-northwesterly and the other northeasterly. Some of the north-northwesterly fractures have fault offset, down on the west; original movement was pre-ore, but re-activation occurred during Basin and Range tectonics.

The ore consists of argentite, native silver, and perhaps cerargyrite in jasperized Guilmette Limestone. The silica and silver appear to be contemporaneous, and are slightly older than the mid-Tertiary intrusive rhyolite. The solutions entered along the fracture systems and deposited the minerals in crakle breccia in and near the axis of the anticline, at or near the Guilmette-Pilot contact. A period of calcite deposition followed intrusive body, though evidence of the silver may have been a deep-seated sources were the Chainman Shale and the mid-Tertiary intermediate flow rocks. Some supergene redistribution has occurred, giving the deposit its present uniform, blanket-like form. The deposit occupies about 70 acres; averages deposit crops out in many places). The deposit contains seven million tons averaging three ounces silver per ton, and is haloed by a like tonpage of lower

Ore was discovered in the district in 1868, and about 100,000 tons averaging 20 ounces silver per ton were mined before 1885. Silver King Mines explored and developed the district in the 1960's and 1970's and is presently constructing a 1200 tpd, CCD cyanide plant scheduled to go on stream in early 1981.

GEOLOGY AND ORE DEPOSITS OF THE TAYLOR SILVER DISTRICT

LOCATION

The Taylor Silver District is located near the eastern border of Nevada, in White Pine County (Fig. 1). The district is 14 miles south of Ely, and four miles east of Highway 6, 50, and 93.

Silver King Mines, Inc., and Agnew Enterprises are joint-venture partners in the property, with Silver King as operator. The companies own four patente mining claims in the district; eight claims for which patent proceedings are underway; and 108 unpatented claims, all in Humboldt National Forest.

GEOLOGY

General'

The district lies on the western slope of the Schell Creek Range, at elevation 7500 feet. Rocks in the range consist of Paleozoic sediments, mainly limestone, dolomite, and shale. The sedimentary rocks were intruded by a few small bodies of mid-Tertiary rhyolite. Remnants of mid-to late-Tertiary intermediate flows and pyroclastic rocks occupy the western foothills of the range.

Stratigraphy

The oldest formation exposed in the central Taylor district is the Guilmette Limestone, Devonian in age (Fig. 2). The Guilmette is about 2,000 feet thick. It consists of massive, fine-grained limestone in the upper few hundred feet, grading downward into sandy limestone and sandy dolomite. The lower contact is gradational into the Devonian Simonson Dolomite. The Guilmett is resistant to exosion, especially in its upper part, and forms massive cliffs

The Guilmette is overlain by a 100 foot thick section of thin-bedded limestone and siliceous shale. This section represents a transition between the Guilmette and the overlying Pilot Shale.

The Pilot Shale is Devonian and Mississippian in age, and consists of 300 feet of banded, siliceous shale. The Pilot forms distinctive tan-weathering talua slopes.

The Joana Limestone disconformably overlies the Pilot Shale. The Joana is about 300 feet thick. It is a typical Mississippian limestone consisting primarily of thick-bedded to massive coarse-grained limestone which is composed mainly of crinoid fragments and other fossil remnants. The upper one-third and lower one-third of the Joana forms massive, monument-like cliffs in eastern Nevada. The central one-third is thinner-bedded, less resistant, and forms talus slopes.

The Mississippian Chainman Shale disconformably overlies the Joana Limestone. The Chainman consists of incompetent, black, euxinic shale; the formation typically forms strike valleys which exhibit chaotic, landslide topography where younger rocks have slid over the chainman. Numerous sandstone and quartzité beds, mainly in the upper part of the formation, form small ridges.

The depositional thickness of the Chainman is difficult to determine, because of poor exposures and because of the tendency of the Chainman to fold or flow under stress, thus causing missing or repeating sections. Sadlick (measured 23 sections in Eastern Nevada and Western Utah, and estimates the Chainman was in excess of 2,000 feet thick, In the Taylor district, the maximum observed thickness is about 1,000 feet. Ten miles to the west, is the Ward Mining District, the average thickness in 80 core holes is 600 feet.

The Pennsylvanian Ely Limestone overlies the Chainman Shale. The Ely consists of more than 2,000 feet of cyclicly deposited thin-bedded limestone, shaly limestone, and calcareous shale. The limestone beds are predominately fine-grained, though a few beds are composed of crinoid and brachiopod fragments.

It is interesting to note the number of places in which this section of the stratigraphic column (uppermost Guilmette to basal Ely) is the locus for mineral deposits in Eastern Nevada.

The top of the Guilmette contains silver ore at Taylor, Hamilton (Treasure Hill bonanza mined in the 1870's), and several other smaller occurrances; these beds are also the host for the very extensive copper-zinc-silver deposits in the Mard district.

The transition beds above the Guilmette are the host for the gold deposits at Alligator Ridge; these beds and the overlying Pilot Shale are under intensive exploration in eastern Nevada at present.

The Joans limestone is the host for silver mineralization at Taylor and Hamilton; silver-gold ore in deposits which are satellite to the Ruth copper porphyry deposits; and extensive zinc-silver-copper deposits at Ward.

The Chainman Shale and Ely Limestone are the host rock for about 20% of the copper ore mined at Ruth (the remainder of the ore is in quartz monzonite porphyry). The Ely is the host for lead-silver-zinc deposits at Ward.

Intrusive activity was confined to the mid-Tertiary (35 M.Y.±) a few rhyolite dikes, sills and irregular bodies intruded the sedimentary rocks. The rhyolite bodies contain xenoliths of silver-bearing jasper, but are themselves intensely hydrothermally altered to clay minerals. We infer that the rhyolite was emplaced only slightly later than the silver and silica.

Structure

The Schell Creek Range is a north-trending, eastward tilted horst typical of the Basin and Range geomorphic province. Rocks in the range have been subjected to three distinct periods of diastrophism.

1) Mid-Mesozoic compressional phase ("antler" or related movement) which resulted in north-trending tight folding and decollement thrust faulting, and extensive deformation of incompetent units such as the Chainman Shale. These stresses in the Taylor District formed a north-trending, assymetrical anticline (Fig. 4); the competent Guilmette limestone fractured to crackle breccia on the crest and flanks of the anticline; and this breccia became the host for the Taylor ore deposits.

2) "Daramide" uplift and intrusive phase, which extended intermittently from 120 to 35 million years ago. Numerous north to north-northwest trending, high-angle, predominantly normal faults with small displacement were first activated in the Taylor district during this phase. A complementary east-northeast set also developed at this time. These faults were the conduits for the hydrothermal silver-bearing fluids which deposited the silver ore in the Guilmette crackle breccia.

Notably few igneous bodies were intruded into rocks of the Schell Creek Range during this phase. They are limited to a few rhyolite plugs and dikes in the vicinity of the Taylor district. Near the close of this diastrophic phase, the intermediate flows and pyroclastics which now occupy the foothills of the range were extruded.

3) Late Tertiary to recent "Basin and Range" phase. This continuing orogeny has formed the present geomorphology of Eastern Nevada. Total structural relief from range to valley is typically several thousand feet. Many of the north-trending faults in the Taylor district were re-activated during this period.

ORE DEPOSITS

Mineralogy

The ore in the Taylor district consists of finely disseminated crystals of argentite and clots of native silver in a gangue of silicified limestone which has described as jasperoid. Much of the jasperoid consists of breccia fragments. Accessory minerals include limonite pseudomorphing pyrite, and calcite and quartz as late-stage veins and as the matrix cementing the jasperoid breccia.

Lovering has identified other sulfide minerals such as stibnite, sphalerite, tetrahedrite, chalcopyrite, galena, and pyrargyrite (?), but these are rare and are of no economic importance.

Source of Silver

Both Drawes and Lovering relate the argentian hydrothermal solutions to a deep-seated igneous body which they believe was emplaced under the Taylor district during Cretaceous or Tertiary time.

The existence of such a stock is open to question in my mind. The Schell Creek Range is distinctive for its paucity of intrusive bodies or thermal effects related to such bodies, and aeromagnetic surveys flown at various elevations over the district reveal no anomalous magnetism which might reflect a deeply buried intrusion.

Another possible source for the silver (and the silica) is the Chainman Shale. Robert Boyle and others who have sampled black euxinic shales such as the Chainman, report silver content of one to ten ppm. If the Chainman Shale overlying the Taylor district contained one ppm silver, then each square mile of Chainman would have, upon deposition, contained three times the silver presently contained in the Taylor deposit.

Mesozoic compressional orogeny; because it was much less competent than the overlying and underlying strata, it was folded, deformed, and in many areas has been thinned to a fraction of its original thickness (e.g. East Taylor District). Much heat must have been generated during this period, and the connate water in the Chainman would have dissolved increasing amounts of silver (which is very soluble in water even at standard temperatures - Shcherbina, p. 1/30). Much of this water would have been squeezed from the formation during the mid-Mesozoic and the "Laramide" diastrophism, and would have found its way as a hydrothermal solution into the favorable loci of the jasperoid breccia.

Another possible source of the silver was the mid-Tertiary intermediate flows and pyroclastic rocks which covered the district at one time; the flows may, however, be younger than the silver, as are the intrusive rhyolite bodies.

Supergens Eprichment

Several lines of evidence suggest that the present form of the Taylor deposit has been at least partially determined by supergene redistribution and enrichment of the silver values.

- 1) The deposit is unique in its uniformity; within the ore outline as shown on figure 5, the occurrence of either high grade zones or barren zones is rare, and silver content feathers out both vertically and horizontally.
- 2) A composite of the 130 ore holes in the deposit (Fig. 6) shows a distinct, gradual increase in silver values from the surface downward to a maximum at a depth of about 60 feet, followed by a gradual decline to a depth of 180 feet; at this point, the jasperization of the limestone rapidly decreases as does the silver content.
- The jasperoid near the surface has a leached appearance, being lighter in color and somewhat more "spongy" than the typical jasperoid below a depth of 20 feet.
- 4) A sequential analysis, as follows:
 - A) The rhyolite is younger than the argentian jasper, because fragments of the jasper are common within the rhyolite bodies, having been plucked from the channel walls during emplacement.
 - B) Silver values are typically higher just above rhyolite sills.

 This paragenesis suggests that silver was remobilized after the rhyolite was intruded.

There are several ways by which the primary silver, in the form of argentite and perhaps also as the native metal, could have been dissolved in the zone of oxidation, and re-precipitated at the redox interface. Boyle and Shcherbina, as well as many others, have studied and reported on this process. Their findings are briefly summarized as follows:

- 1) Dissolution of silver minerals in the zone of oxidation
 - a) Oxygenated water will dissolve argentite, forming the soluble silver sulfate.
 - is oxidized first, forming sulfuric acid as well as ferrous sulfate which further oxidized to form ferric sulfate.

 The sulfuric acid and ferric sulfate then attack the argentite and/or native silver accelerating its decomposition to silver sulfate.
 - the carbonate environment, sulfuric acid may react with the carbonate to form a bicarbonate ion, which will react with argentite to form a soluble silver bicarbonate complex ion.
- 2) Transportation of dissolved silver downward in the surface waters probably as silver sulfate as Taylor.
- 3) Precipitation of silver as native metal or as secondary argentite, caused by the following agents:
 - a) A reduction of the acidity of the solution, because of contact with carbonate wall rocks, will cause the silver to precipitate as an unstable oxide which rapidly breaks down to form native silver.
 - b) A reduction in the oxidation potential of the solution because of proximity to the water table, will reduce ferric ion to ferrous ion. The ferrous ion will precipitate native silver, re-oxidizing ferrous ion to ferrication.

- c) Sulfides in the zone of reduction (e.g. pyrite) will cause silver in solution to precipitate, probably as secondary argentite, according to Schurmann's reaction series.
- d) The reducing action of hydrogen sulfide, generated by decaying organic matter, will cause silver in solution to precipitate, probably as argentite.

HISTORY

The Taylor district was discovered in 1872, during the following 20 years, shout 50,000 tons of ore averaging 20 ounces silver per ton was mined, primarily from the Argus shear zone (Fig. 5). This is one of the north-trending normal fault zones which acted as feeders for the mineralizing solutions. Formations are displaced downward on the east, bringing un-mineralized Pilot Shale in contact with the ore-bearing Guilmette; the fault zone marks the eastern limit of surface silver mineralization. In this area a six foot thick bed at the top of the Guilmette, or perhaps a solution breccia zone parallel to bedding, was stoped preferentially. Ore was also stoped from fault zones and from irregular bodies.

The ore from this mining phase was taken by wagon to Steptoe Valley where two smelters treated it. Because of the long haul and the primitive ore processing facilities, the ore cutoff grade was high, about 10 ounces per ton.

The district became idle about 1892.

Around 1920, a cyanide treatment plant was built at Taylor, but there is no recorded production from it.

During the 1930's, the district was revived; additional ore was mined underground in the Argus area and the Monitor area to the west. Also, some ore was mined by open cut methods in the Argus and Monitor areas and on Bishop Hill,

and most of the old dumps from the 19th century mining, which averaged about eight ounces silver per ton, were also processed. Much of the ore during this period was sold as siliceous flux to the Kennecott smelter at McGill. Total production during this period was about 100,000 tons which averaged 10 ounces silver per ton.

About 1960, K. L. Stoker acquired the existing claims in the Taylor district from several individuals and companies, and when Silver King Mines, Inc., was formed in 1961 the claims became Silver King property. Additional claims were staked, to form the present property holding of 104 mining claims totaling about three square miles. Four of the claims are patented, and patent has been applied for on eight others.

Argus shear zone in an attempt to locate deep ore below the old mine workings.

A small pod of high grade ore was discovered, and in 1964 the Taylor shaft

was sunk to a depth of 400 feet (Fig. 5). The pod produced 4,000 tons in 1965

averaging 30 ounces silver per ton. Further exploration drifting and underground drilling failed to discover significant additional reserves and the mine was closed.

Many of the percussion holes drilled in the early 1960's penetrated substantial thicknesses of low grade silver mineralization at or near the surface. The mineralization was too low grade to constitute one at prevailing silver prices; however, when silver prices began to escalate in 1973, Silver King resumed its exploration and development drilling program. Several core holes were drilled 10 to 20 feet from existing hammer-drill holes, to check the validity of the chip samples. The core data confirmed the chip data (core samples assayed, an average, 5% higher than comparable chip samples). This check drilling also demonstrated the unique uniformity of the Taylor ore deposit; mineralization within the exterior ore bondaries is ubiquitous, and there exist few barren or high grade areas (Fig. 5).

The percussion drilling program resumed in 1974, and has continued intermit-

tently to the present. In all, about 450 holes have been drilled in the district since 1962, of which 22 are core holes. The drilling has outlined an area of about 40 acres underlain by ore grade mineralization, which is defined as at least a 30 foot thickness containing at least two ounces silver per ton. Using this cutoff, the ore averages 50 feet in thickness (up to 200 feet thick in a few places), with average overburden of 50 feet (waste to ore ratio is 1.7:1).

Measured diluted reserves using the two ounce cutoff are presently five million tons averaging 3.3 ounces silver per ton. Much of the overburden is low grade mineralized rock; if all overburden which exceeds one ounce silver per ton is "stripped to the mill", diluted reserves become seven million tons averaging 2.8 ounces silver per ton. Inferred reserves based on reasonable gedlogic projection will add at least three million tons of ore to the reserve.

Financing for a concentrator was secured in 1979; construction of a 1200 ton-per-day counter-current decantation cyanide leach plant began immediately, and was completed in early 1981 (Fig. 8). Total expenditure in the district, including drilling, mine development, and mill construction, was slightly less than \$10 million.